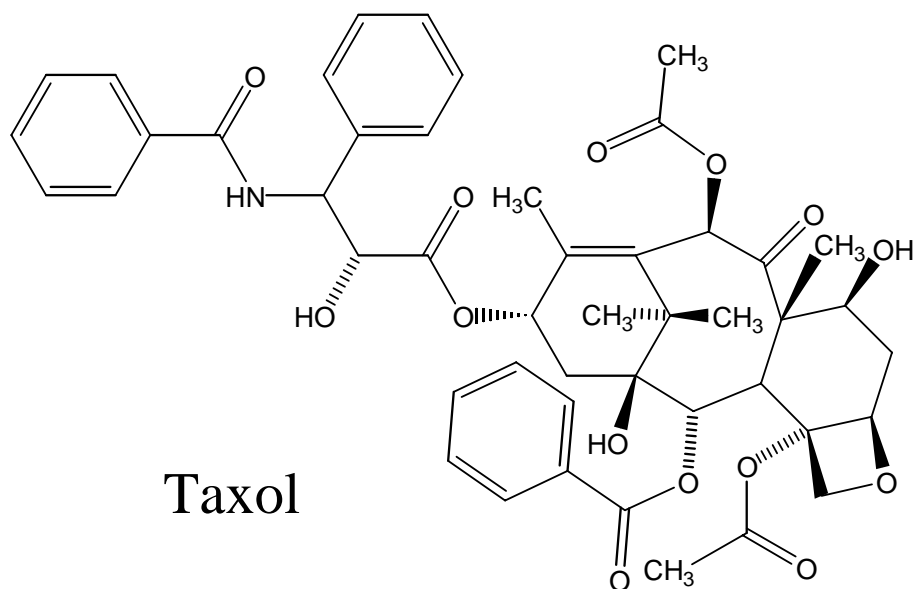


Application of Enzymes and Microorganisms for Organic Synthesis

Jay Keasling

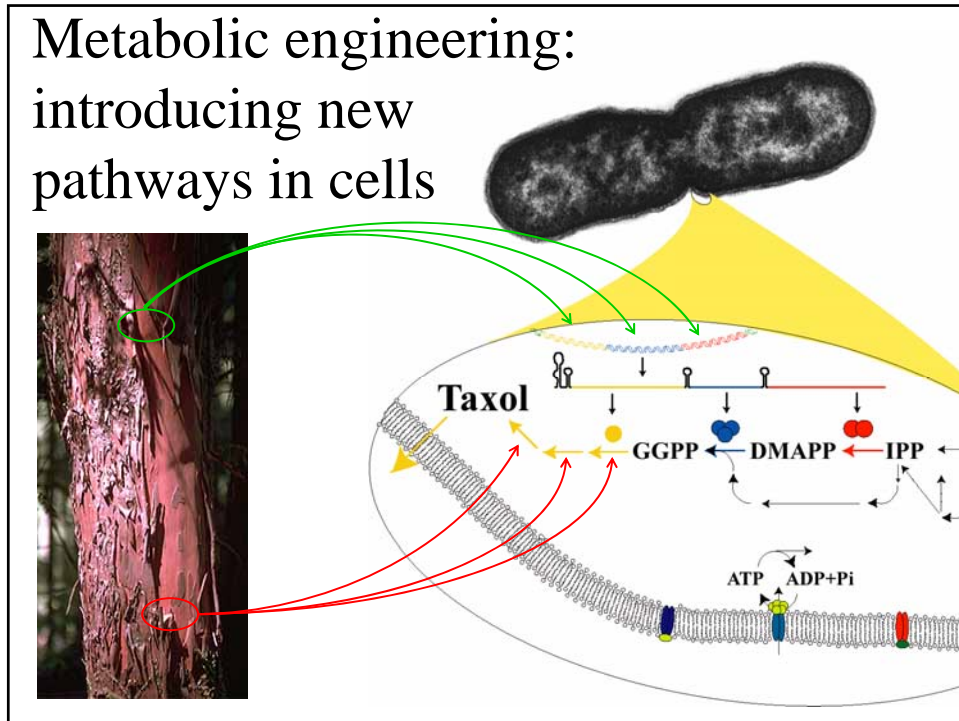


- The Pacific yew tree is an environmentally protected species and one of the slowest growing trees in the world.
- Isolation of the compound, which is contained in the bark, involves killing the tree.
- One 100-year old tree results in approximately 350 mg of taxol, just enough for one dose for a single cancer patient.



Total synthesis of taxol

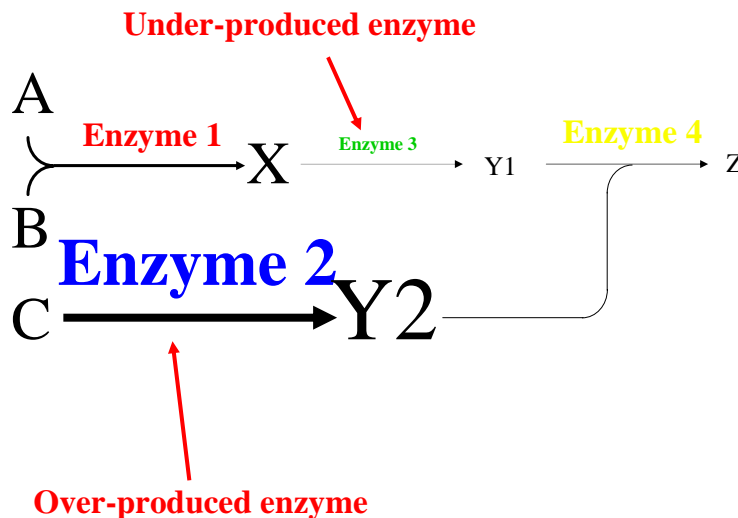
Metabolic engineering: introducing new pathways in cells



Synthesis of complex molecules

- May require several enzymes from one or more organisms
- Expression of genes must be balanced
 - Underexpression of any one gene may limit flux through the pathway and therefore product yields
 - Overexpression will lead to inefficiencies
- Precursors (from inside the cell or supplied from outside the cell) should not severely limit production of the desired product

Balancing enzymatic reactions in the cell

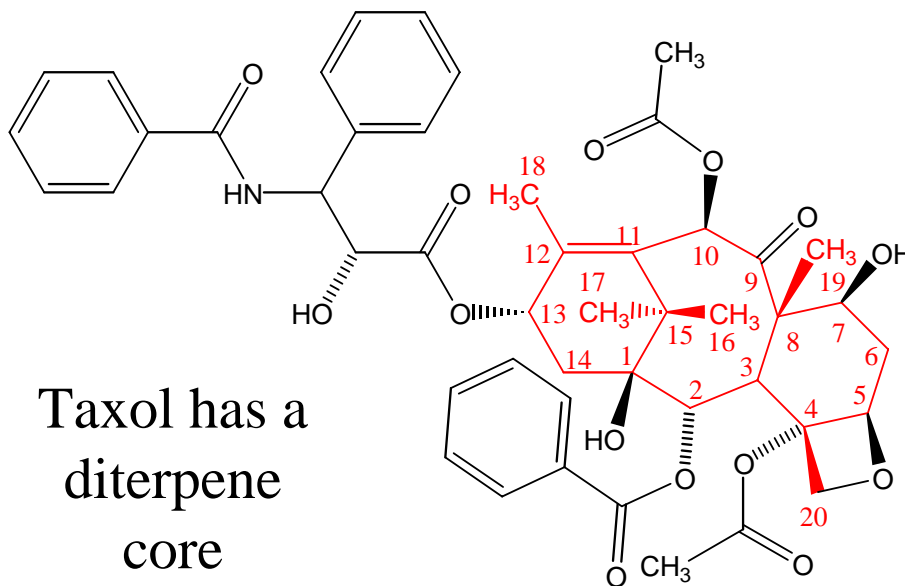


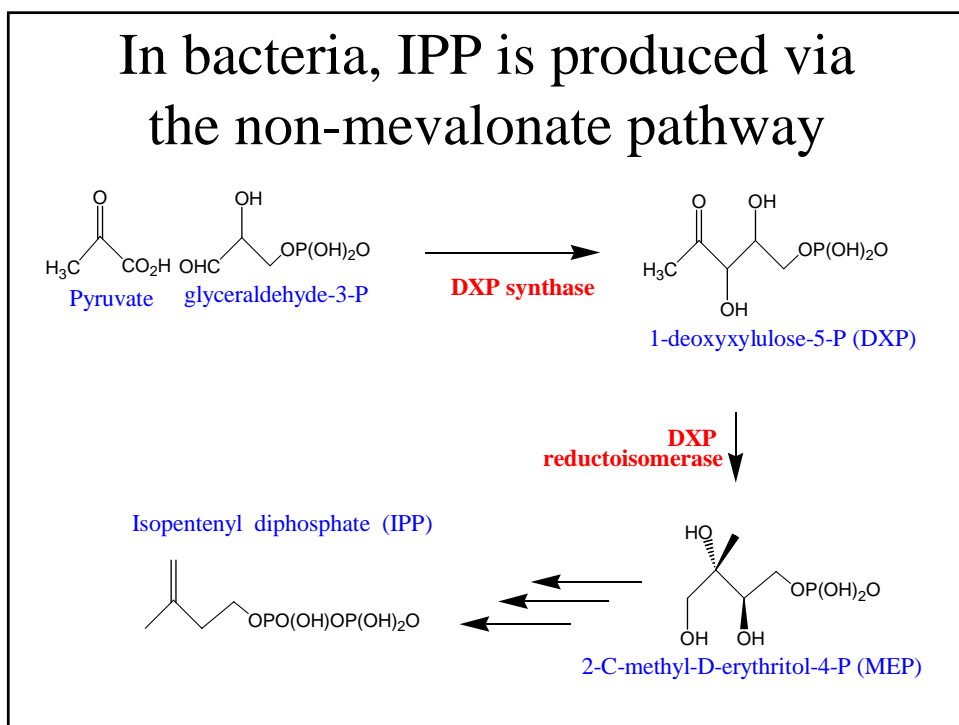
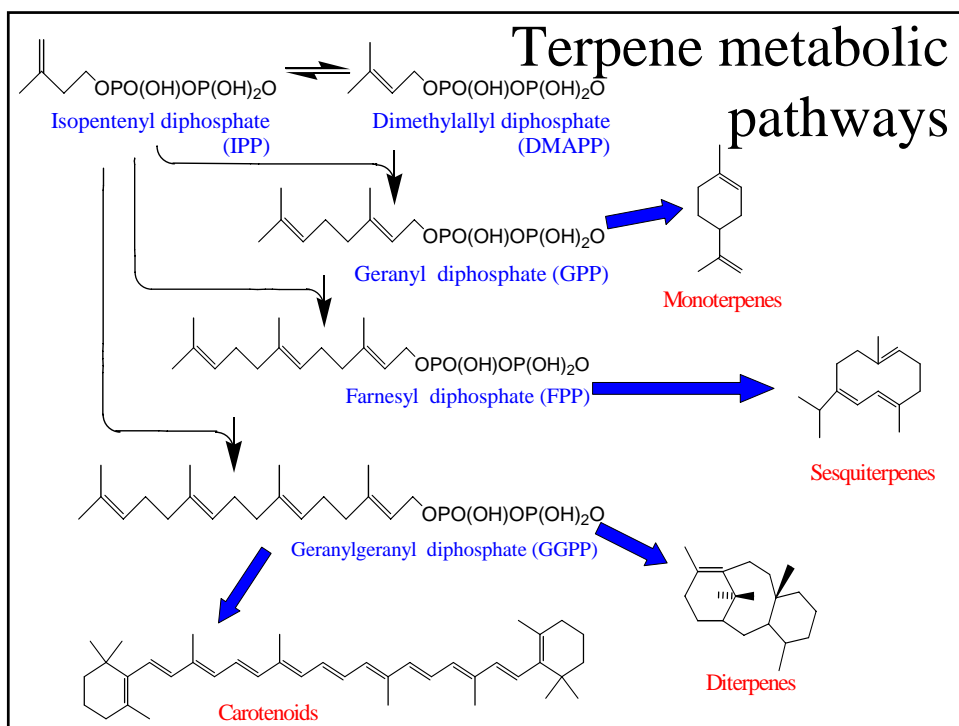
What is metabolic engineering?

- Metabolic engineering is a redirection of enzymatically-catalyzed reactions for the production of a new compound or the degradation of a compound
 - genetic modification of a single organism
 - engineering a consortium of organisms

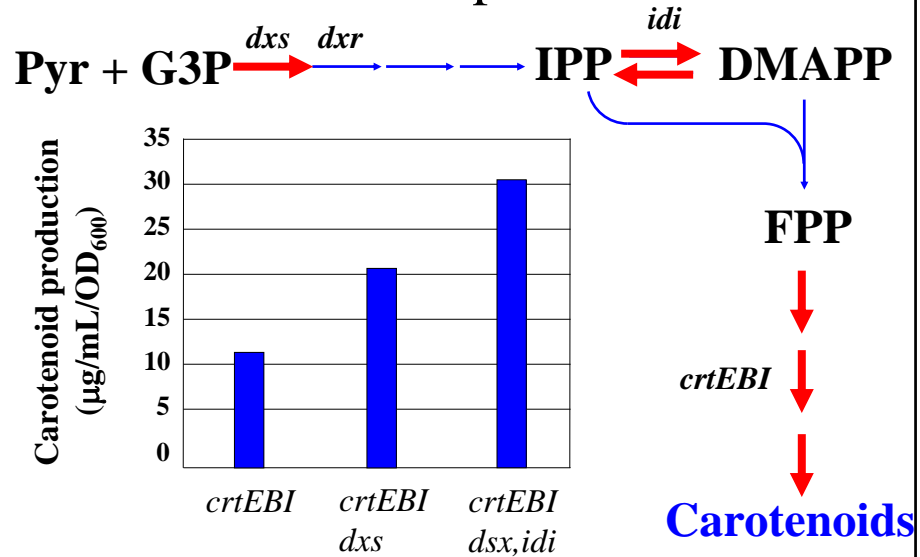
Why do metabolic engineering?

- Production of novel compounds
 - new biopolymers
 - antibiotics
- Production of existing compounds in better ways
- Bioremediation of recalcitrant compounds
 - pesticides/nerve agents
 - PCBs

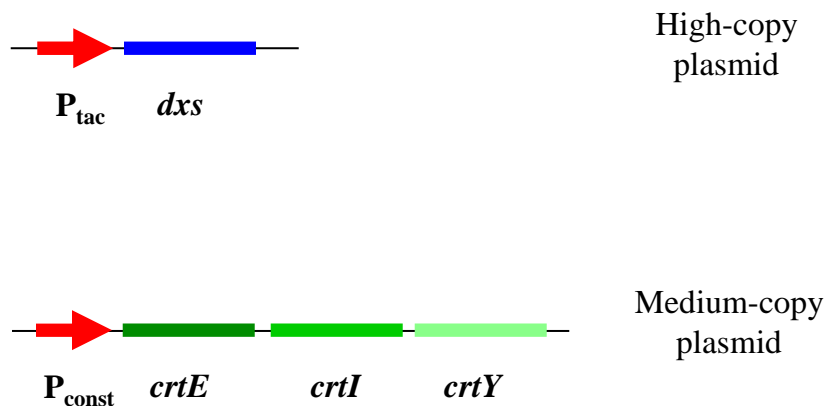




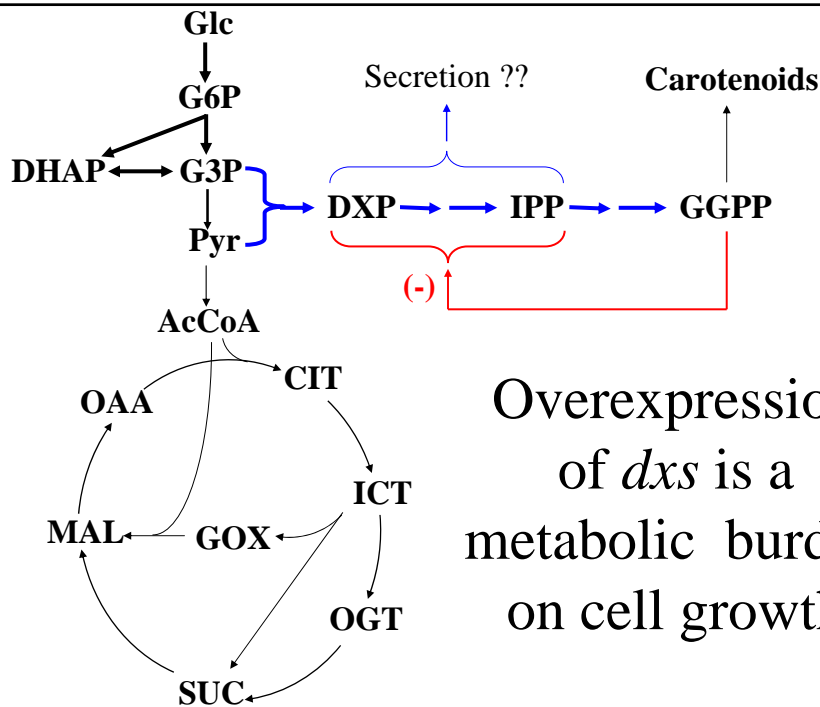
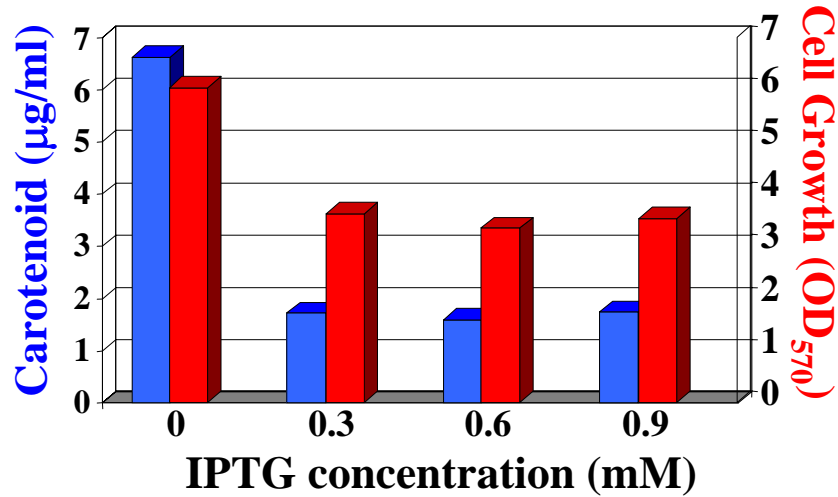
Metabolic engineering of carotenoid production



dxs and *dxr* under P_{tac} control on high-copy cloning vectors



Overexpression of *dxs* from a high-copy plasmid with a strong promoter



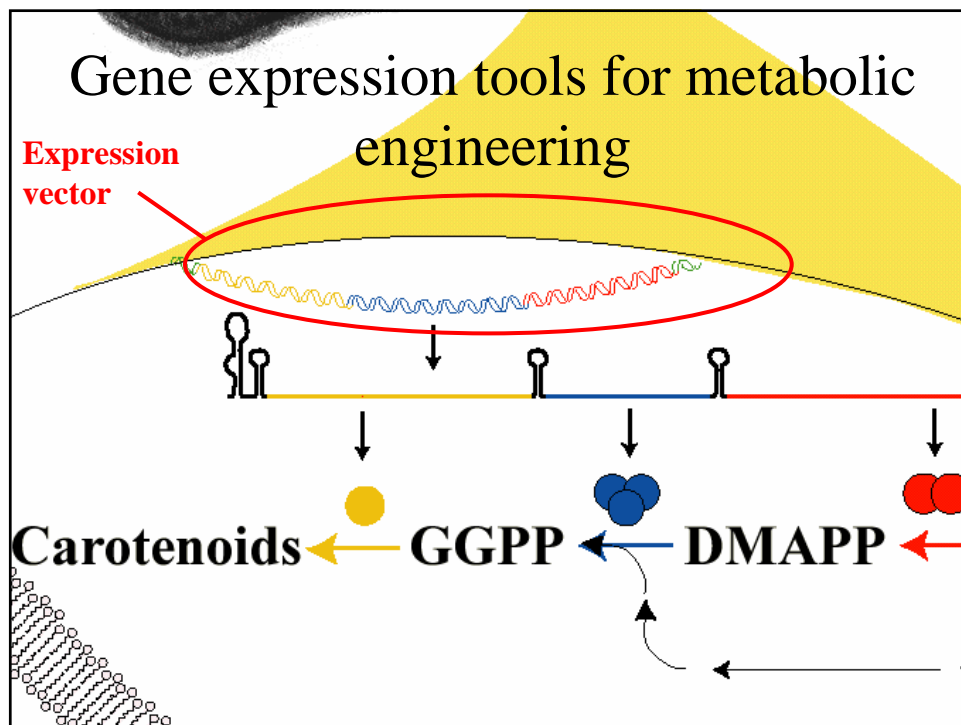
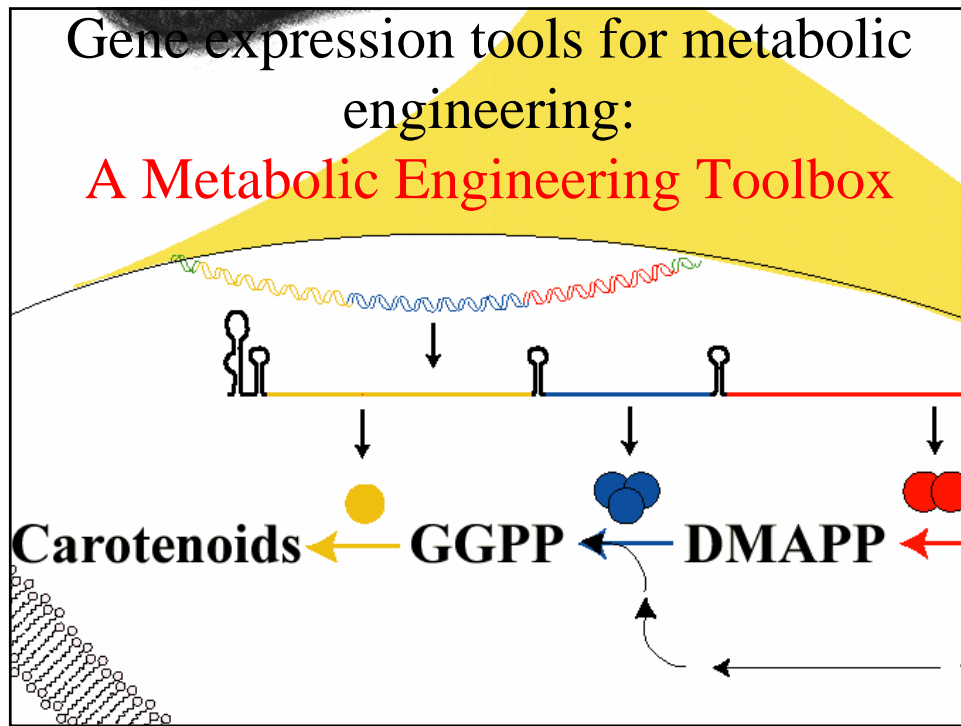
Overexpression of *dxs* is a metabolic burden on cell growth

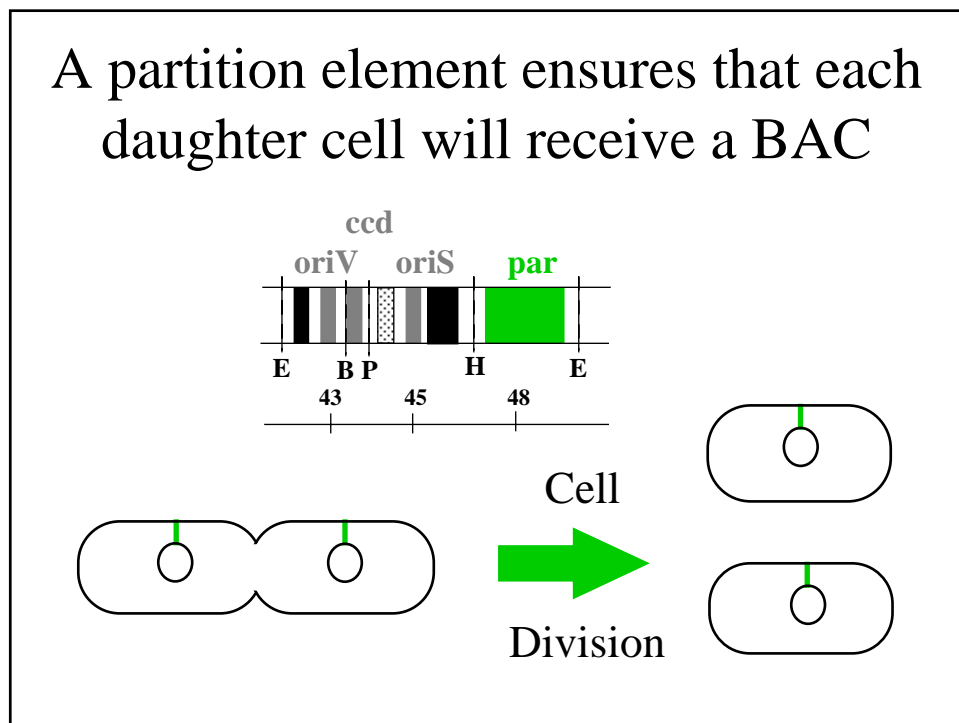
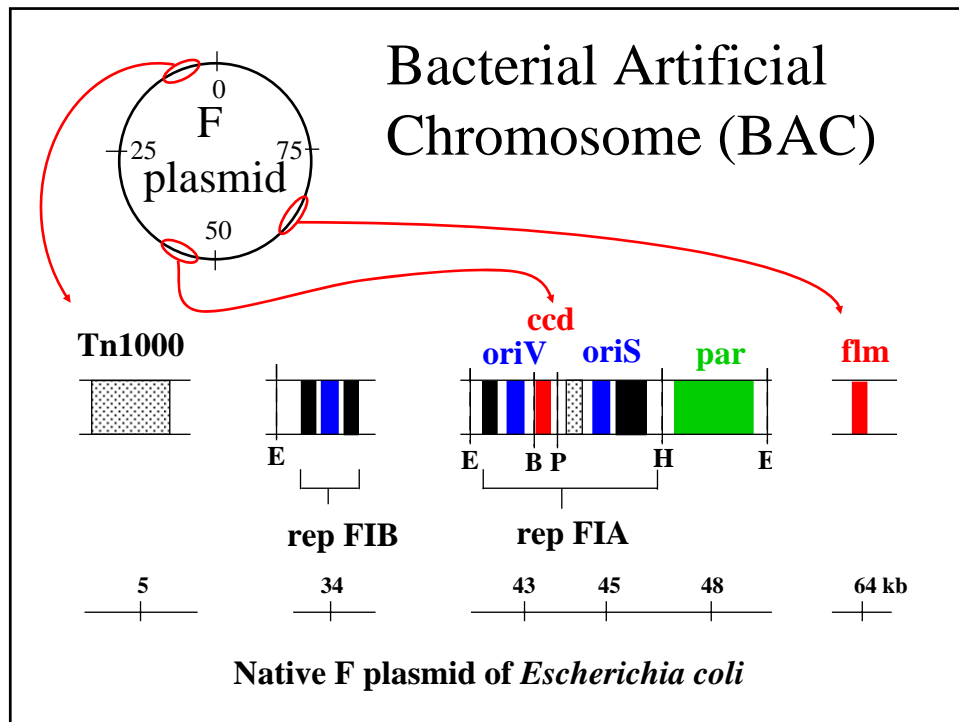
Needs for Metabolic Engineering

- Optimal fluxes through the heterologous metabolic pathways
- Strict control over gene expression
- Consistent control of gene expression in all cells
- Minimal burden of the heterologous genes on the host

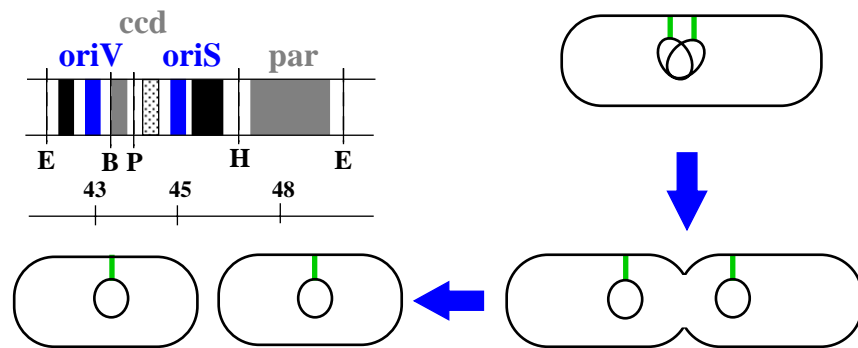
Some key problems in gene expression control

- Many expression vectors are unstable and have variable copy number in the host
- Many promoters do not allow tight and consistent control of gene expression
- There are few techniques to regulate expression of multiple heterologous genes
- Can we predict the levels of gene expression needed for flux redistribution?

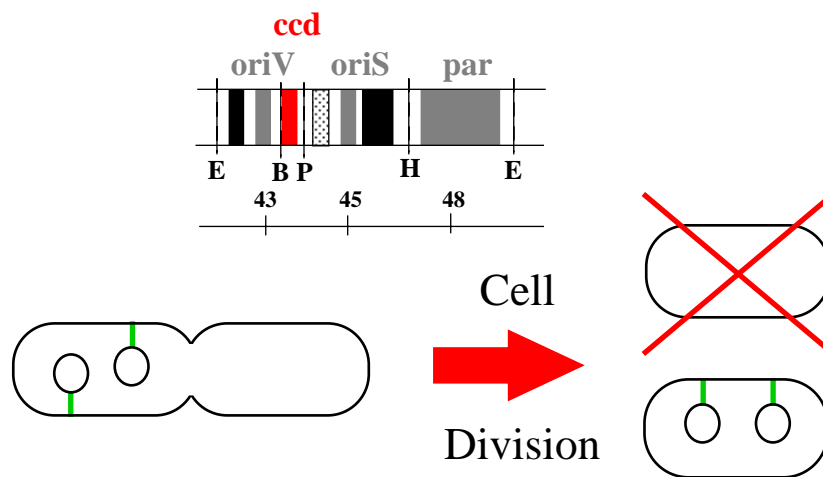




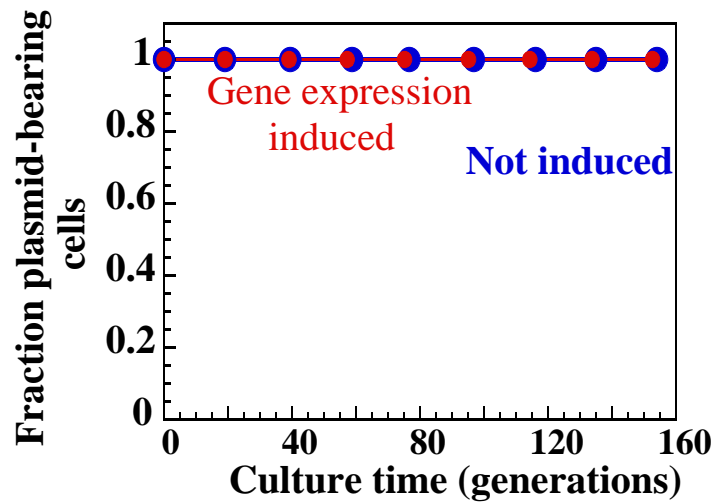
Specific replication origins time BAC replication with the cell cycle



A “Kill” element ensures that BAC-free daughter cells do not survive



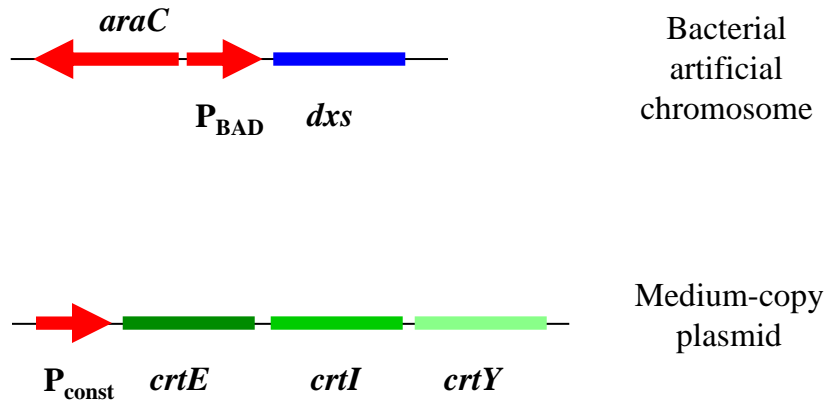
BACs are stable indefinitely in the absence of selection pressure



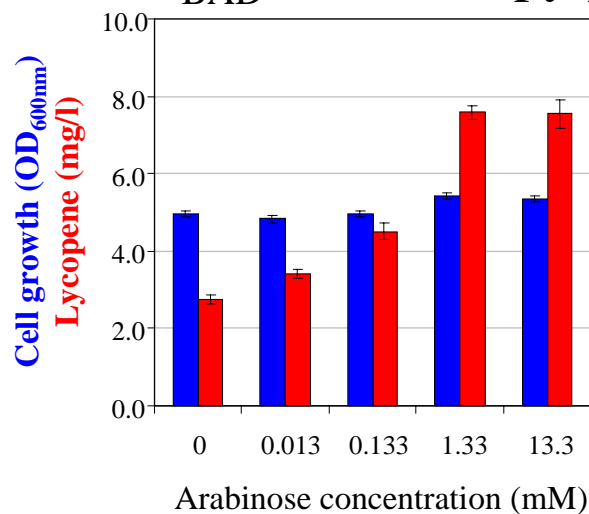
The auxiliary chromosomes have improved control of gene expression

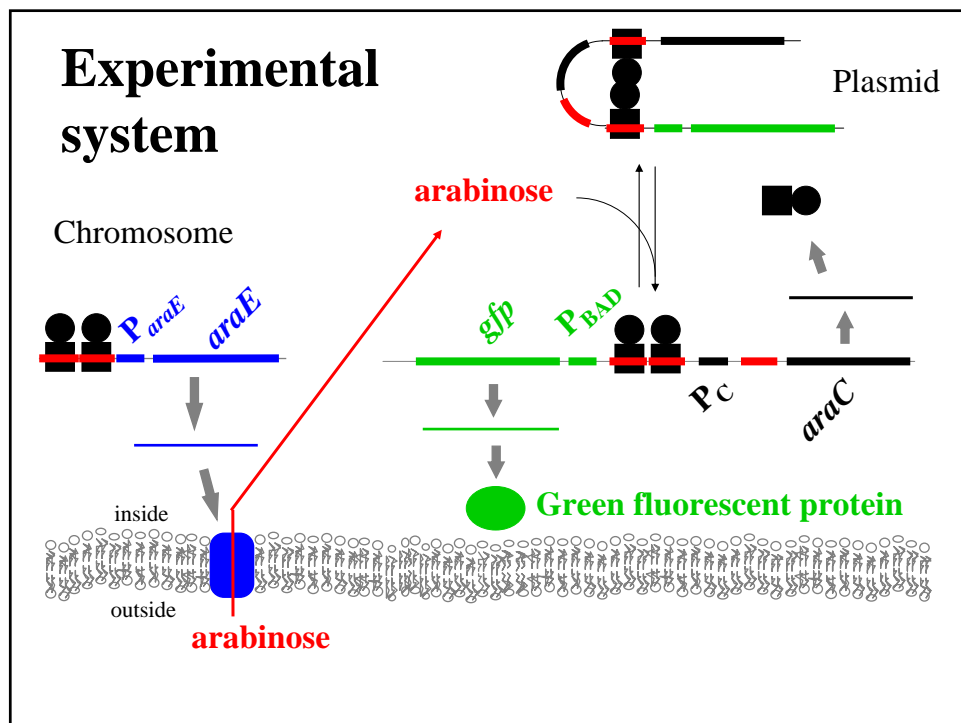
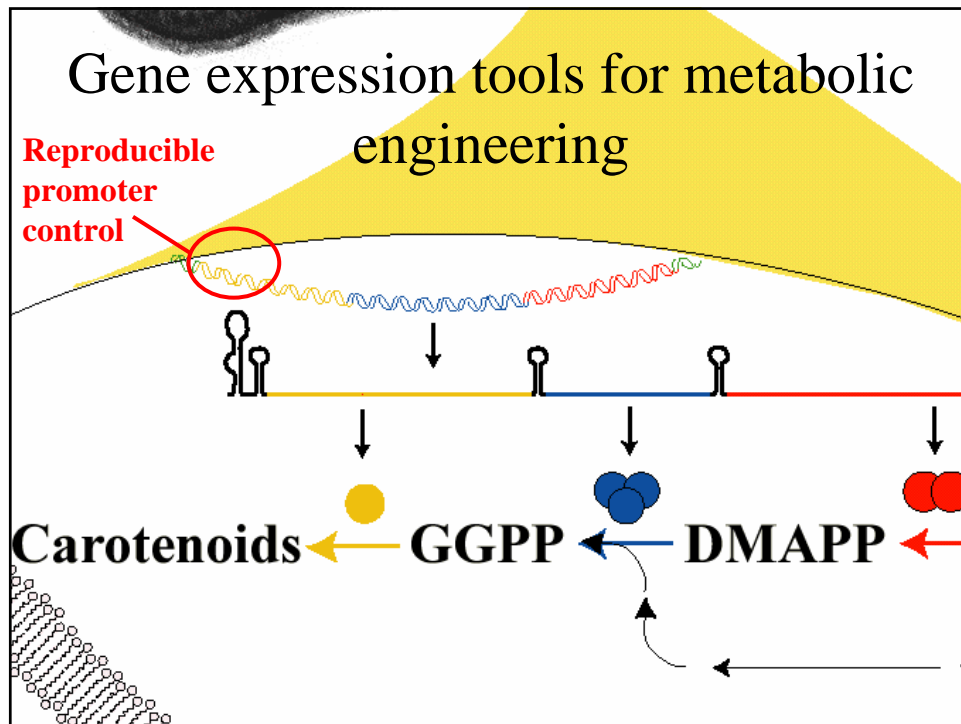
	Uninduced Expression	Induced Expression	Growth Rate of Host
BAC	15 units	4,000 units	0.69 hr ⁻¹
High-Copy Plasmid	200 units	12,500 units	0.53 hr ⁻¹

dxs and *dxr* under P_{BAD} control on bacterial artificial chromosome

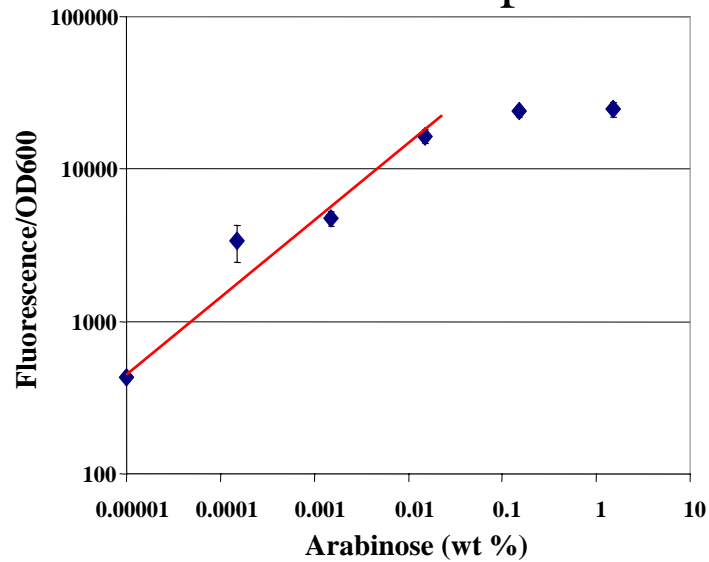


Production of lycopene: *dxs* under control of P_{BAD} on low-copy plasmid

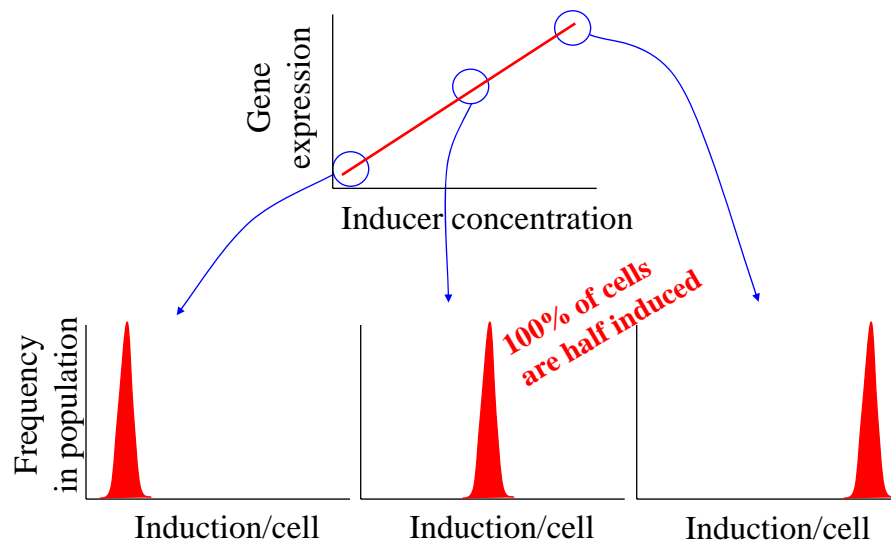




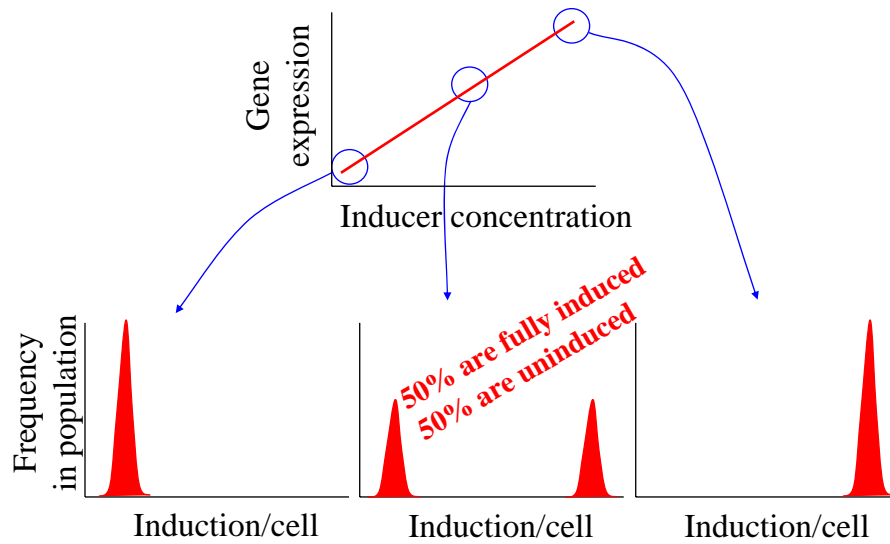
Expression of *gfp* from the arabinose-inducible promoter



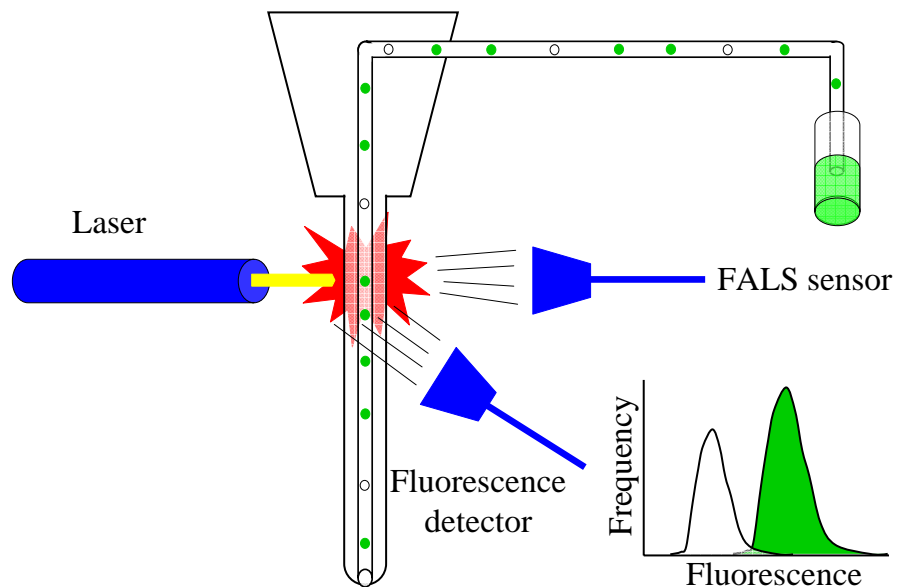
Desired gene expression in population



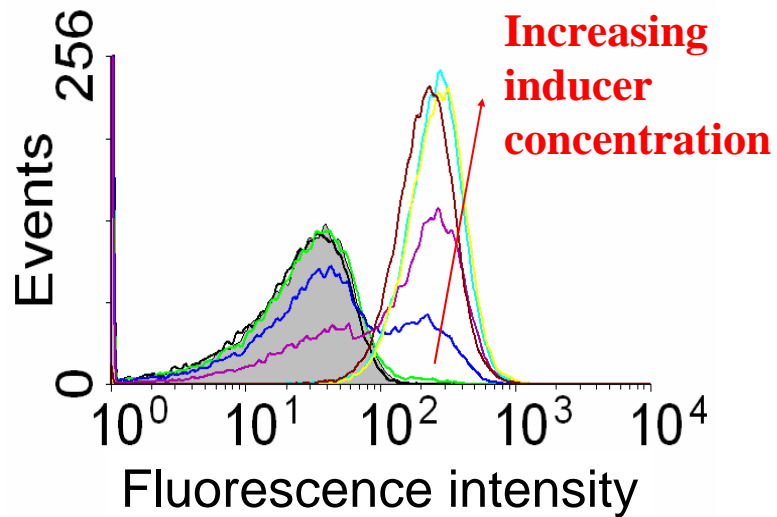
'All-or-None' Gene Expression



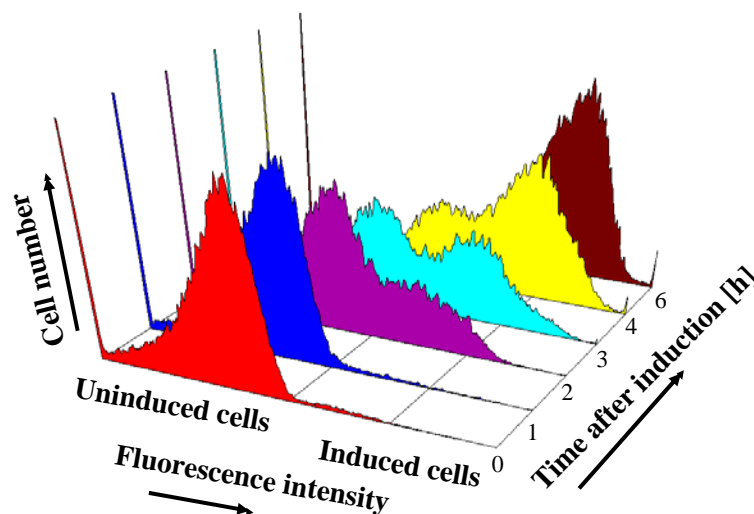
Flow cytometric analysis



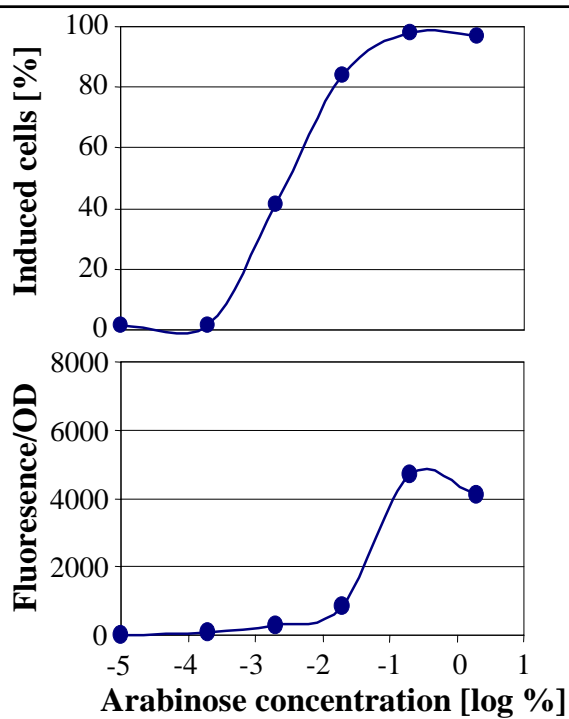
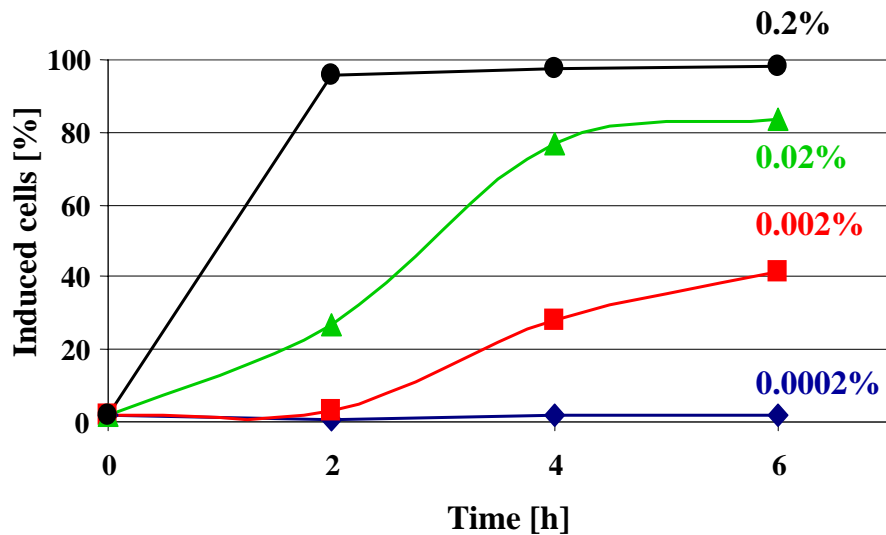
Native arabinose-inducible system
gives rise to two populations



Native arabinose-inducible system
gives rise to two populations

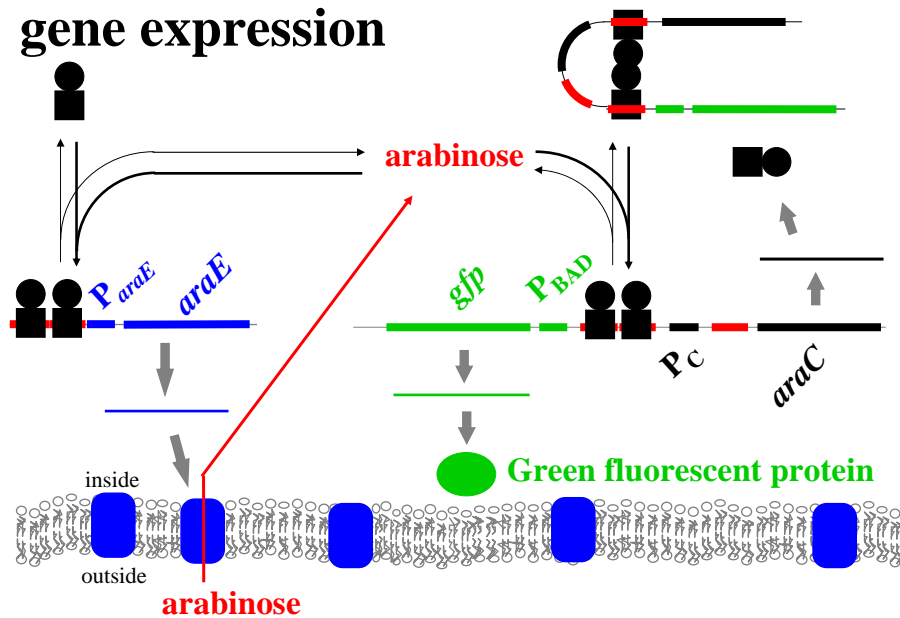


Population dynamics as a function of arabinose concentration

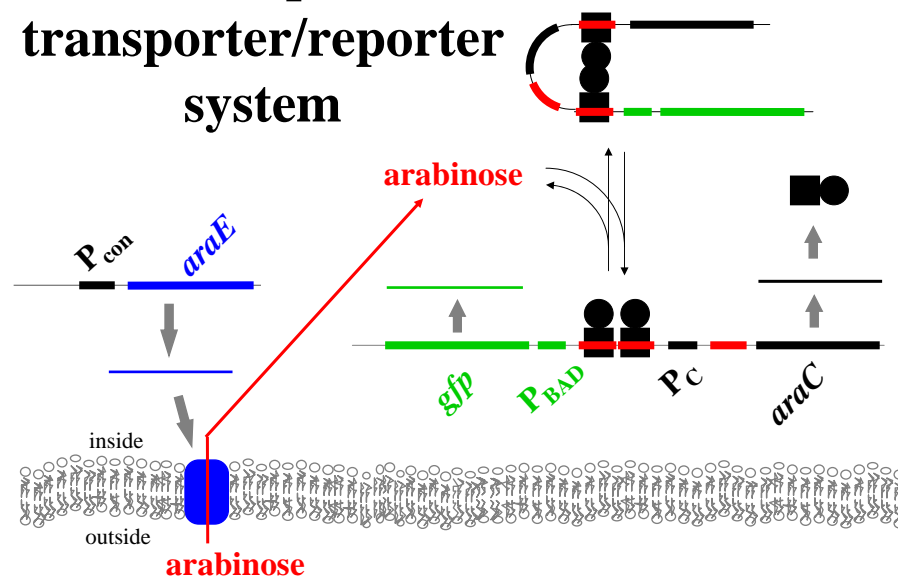


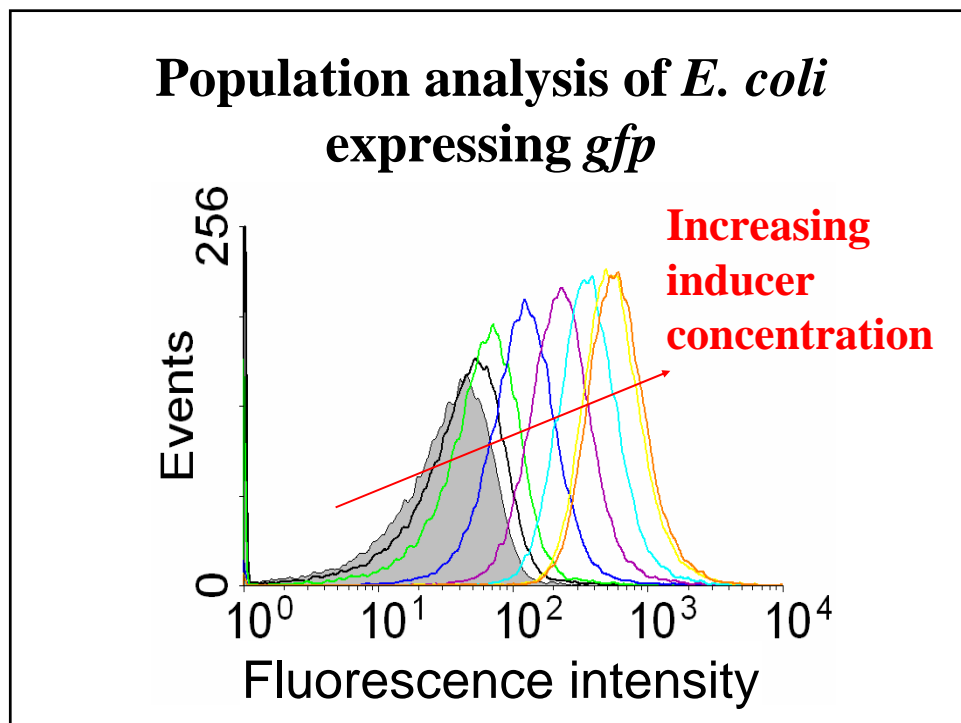
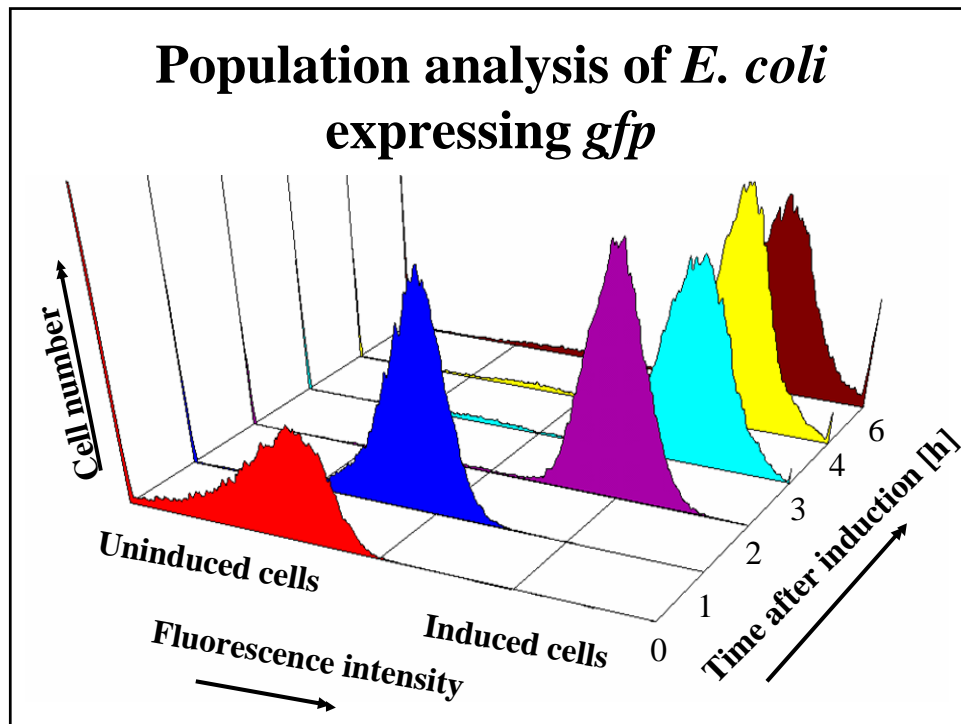
Experiment:
steady-state
induction
levels as a
function of
inducer
concentration

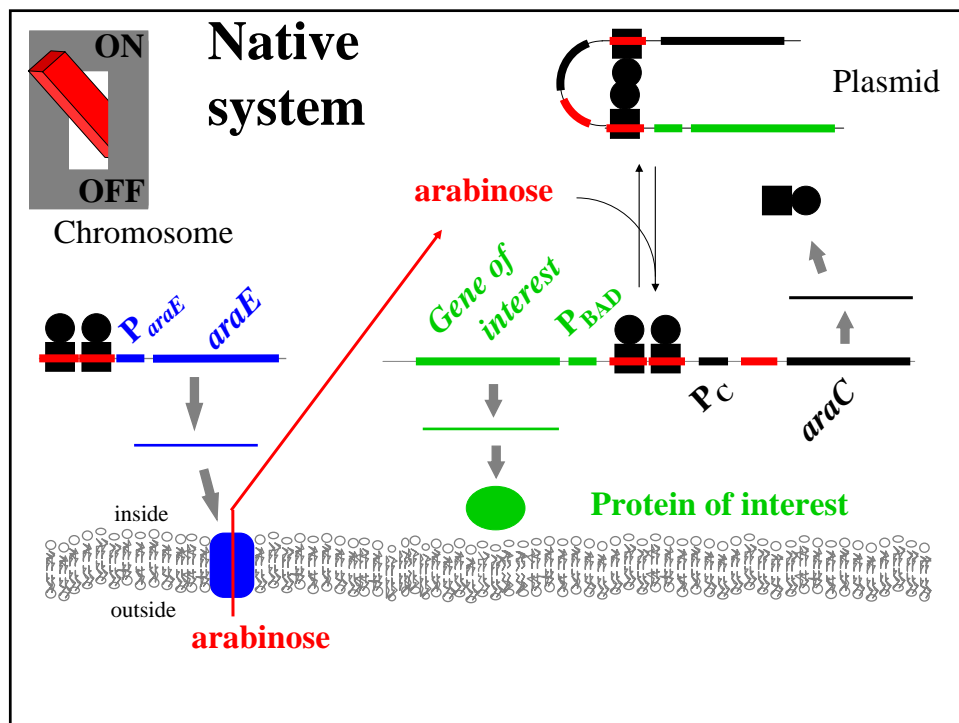
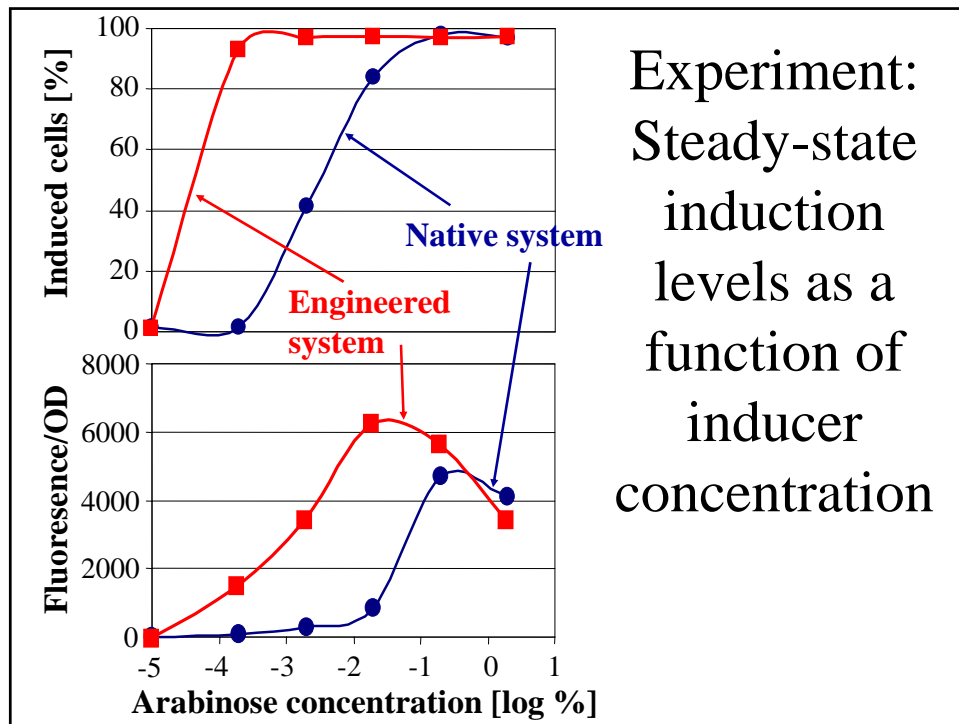
Stimulation of inducer transport gene expression

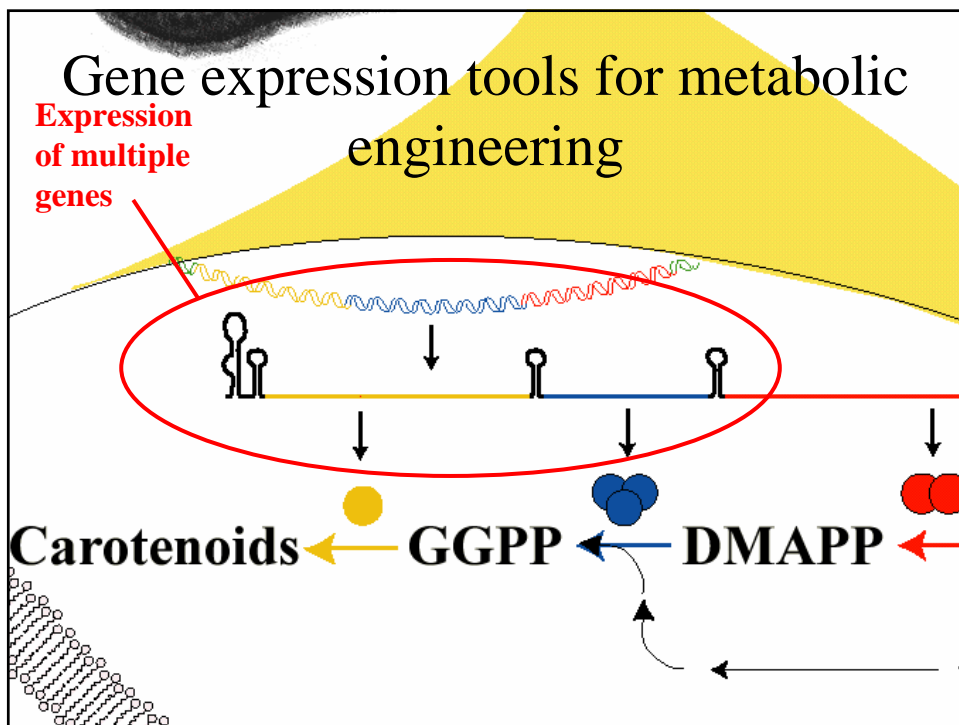
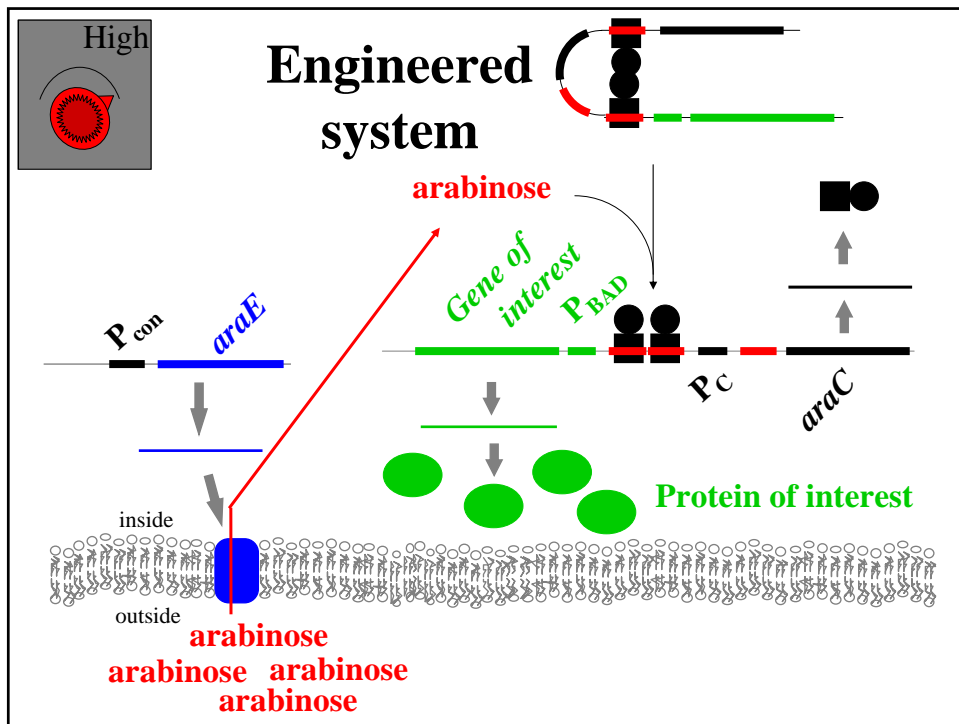


Decoupled transporter/reporter system

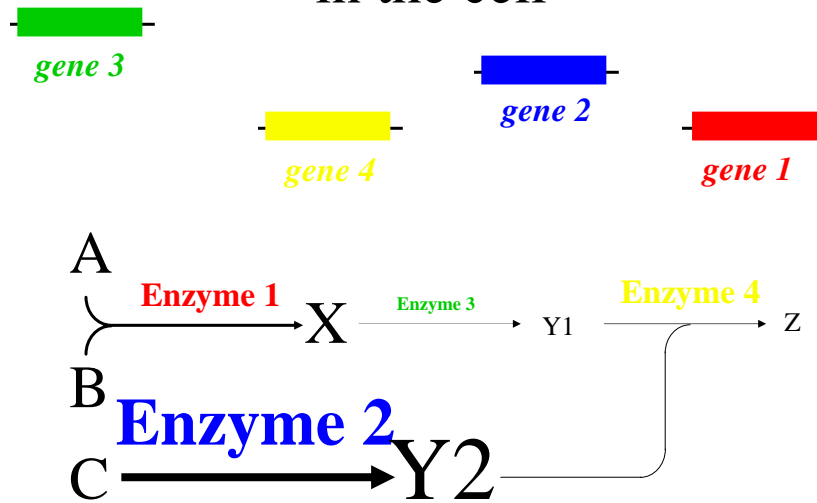




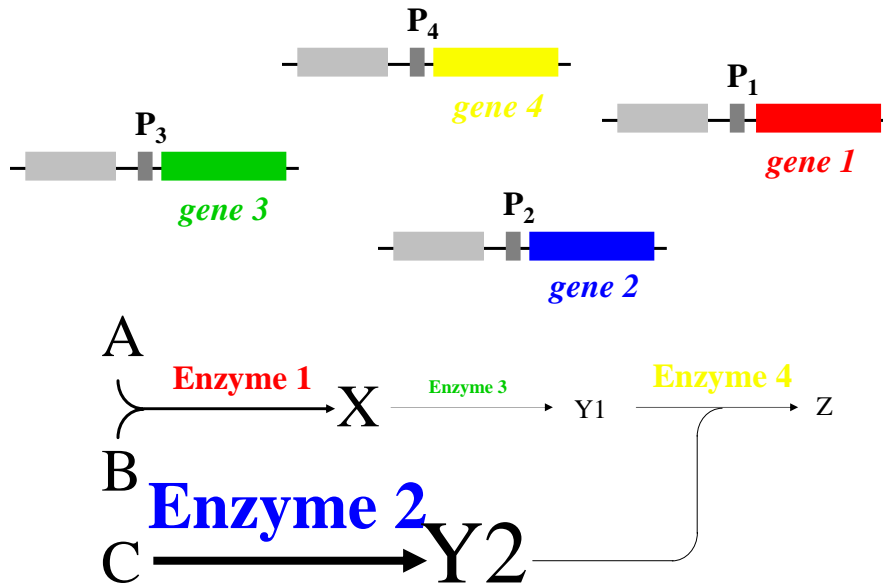




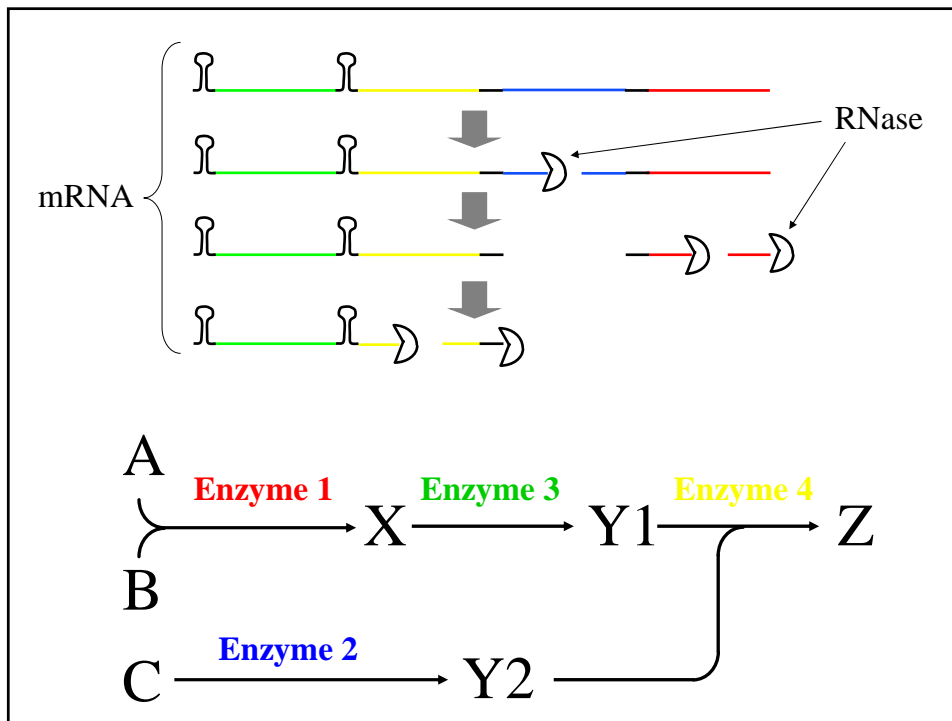
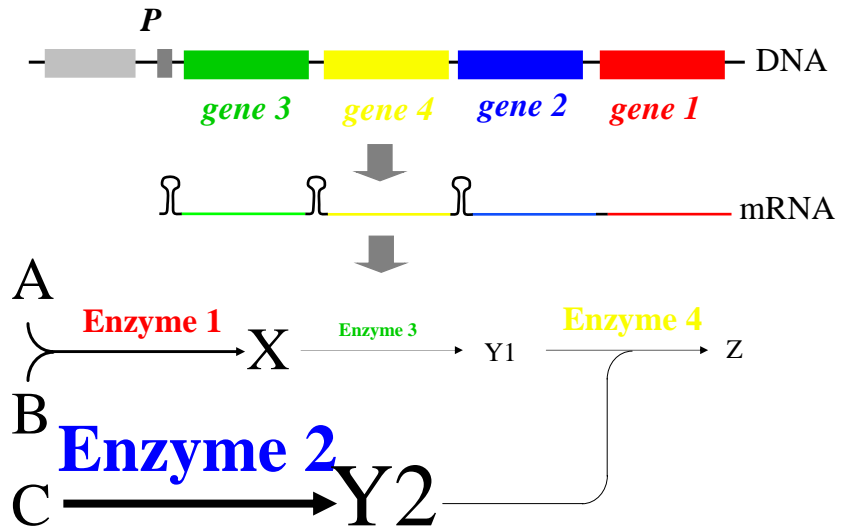
Balancing enzymatic reactions in the cell



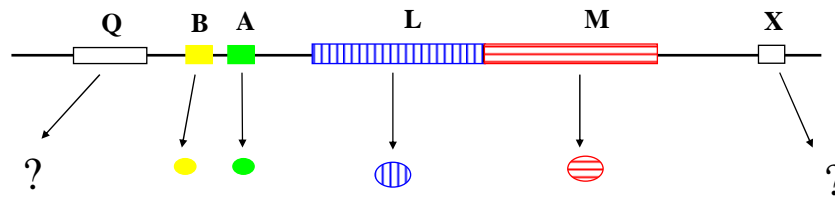
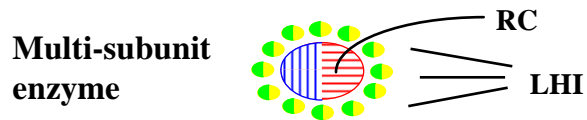
Using individual control elements



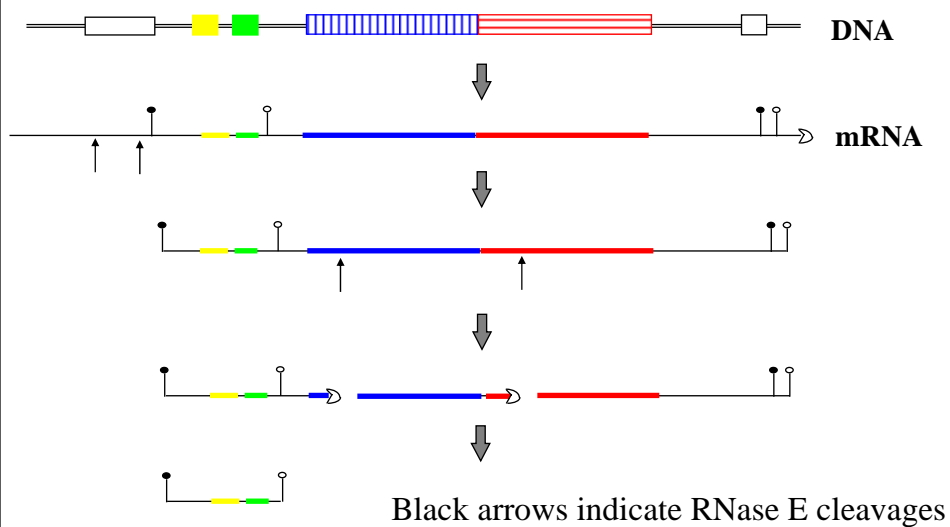
Synthetic operons

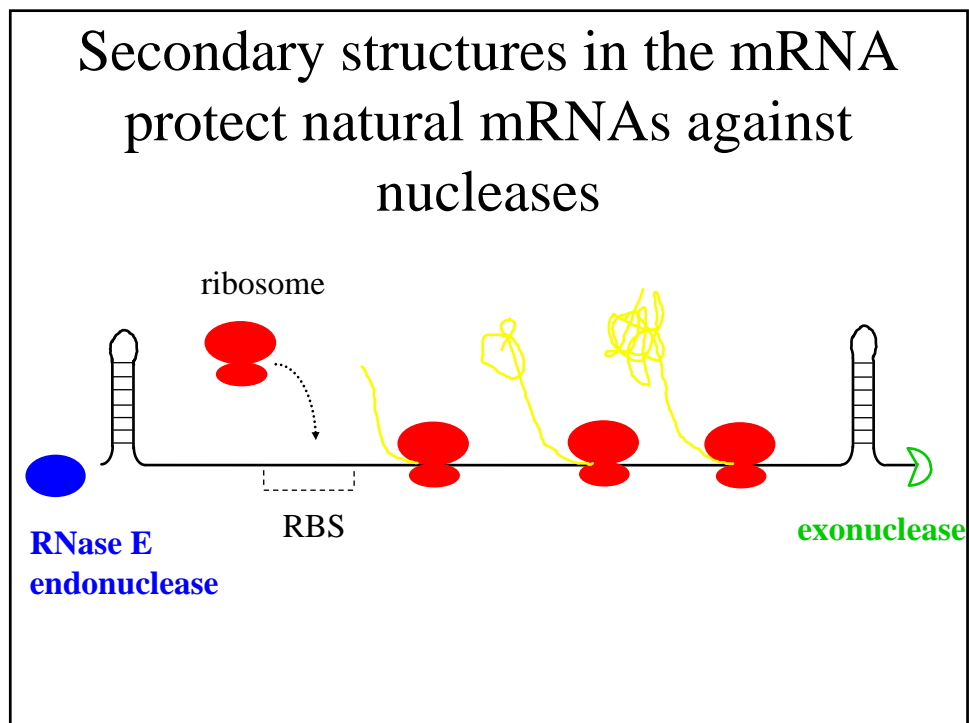
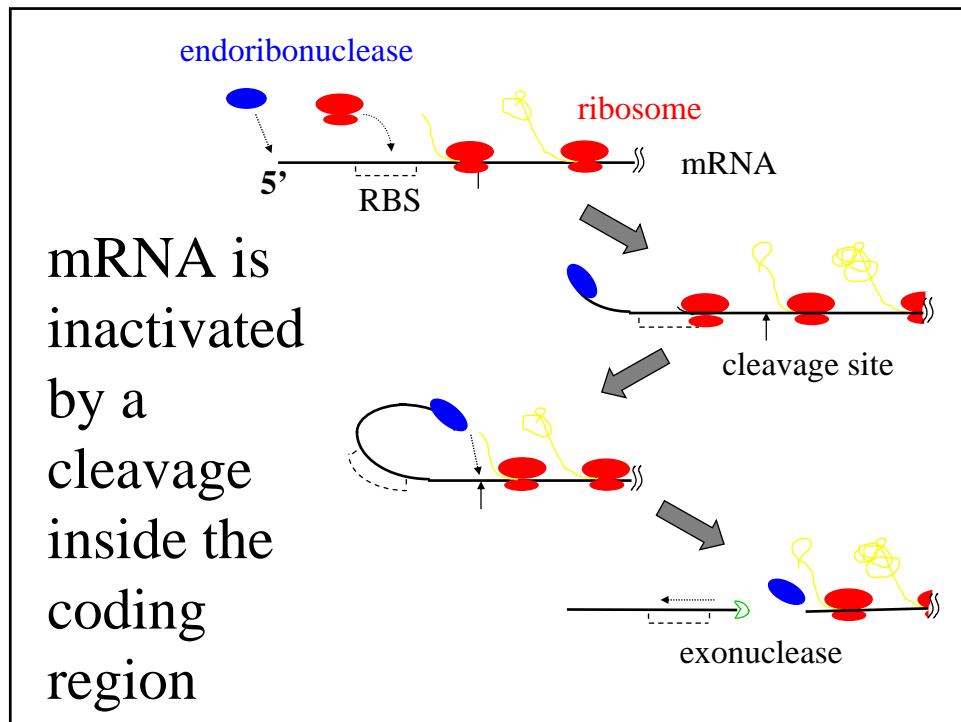


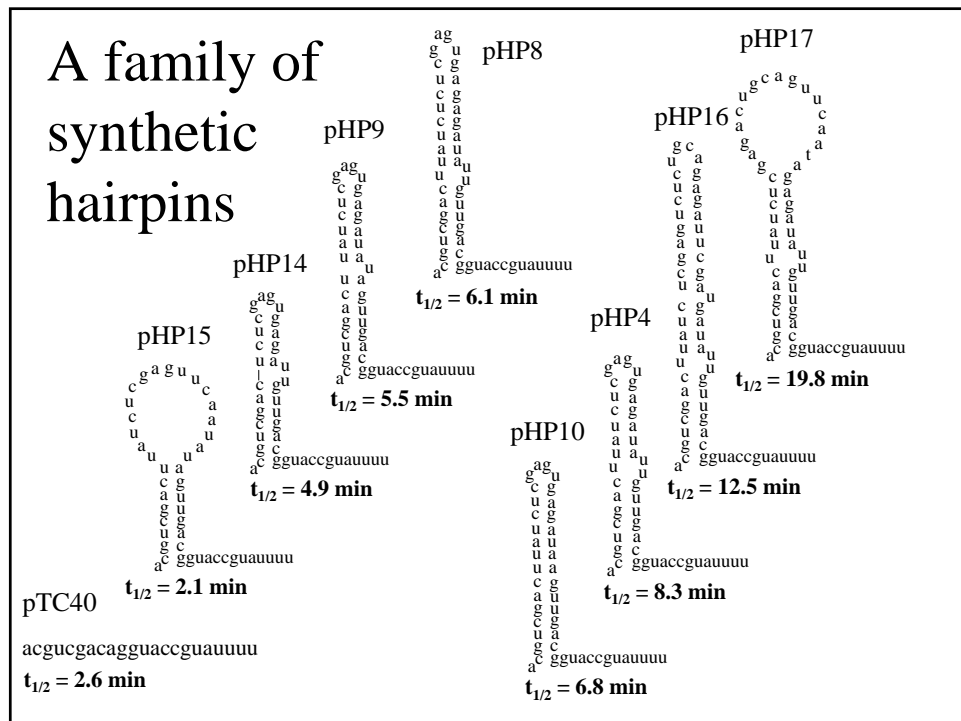
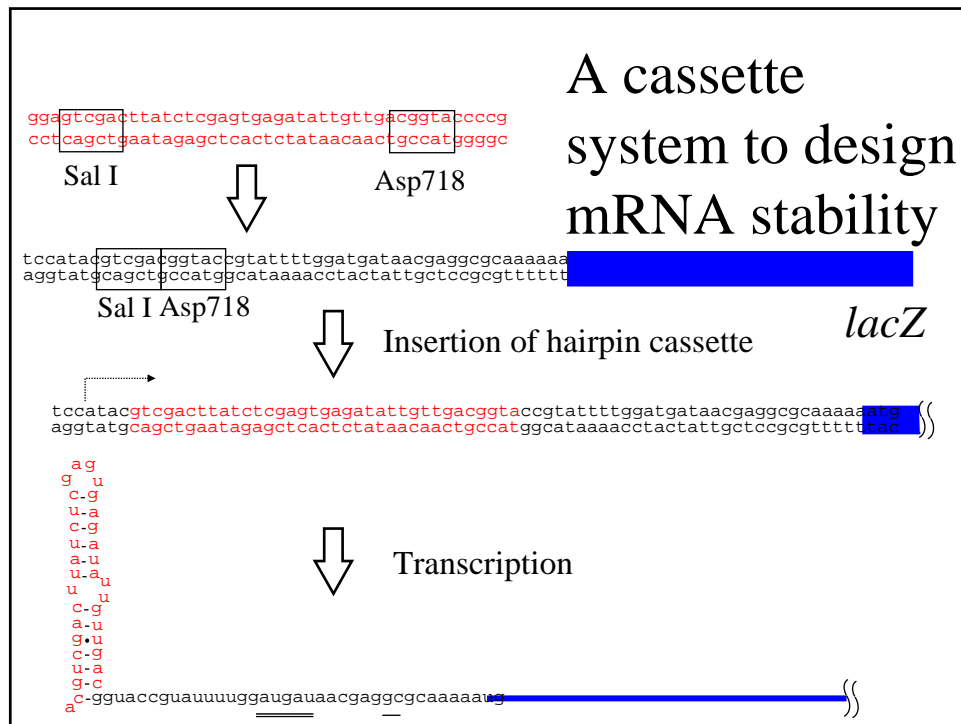
The *puf* operon in *Rhodobacter* encodes a multi-subunit enzyme



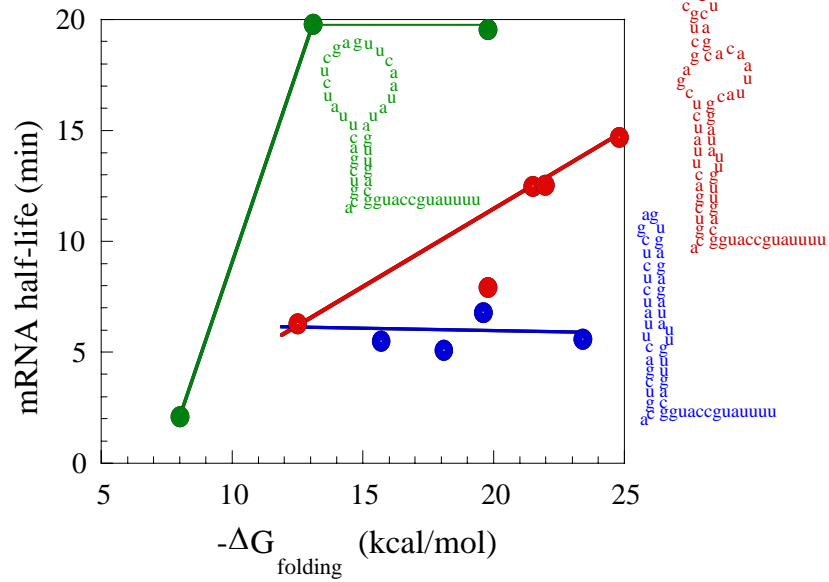
The production of enzyme subunits is controlled by mRNA stability



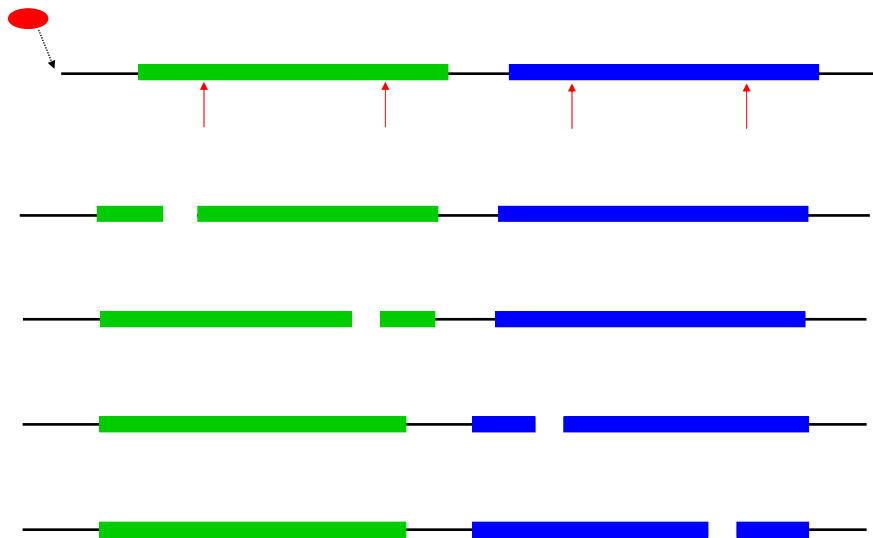




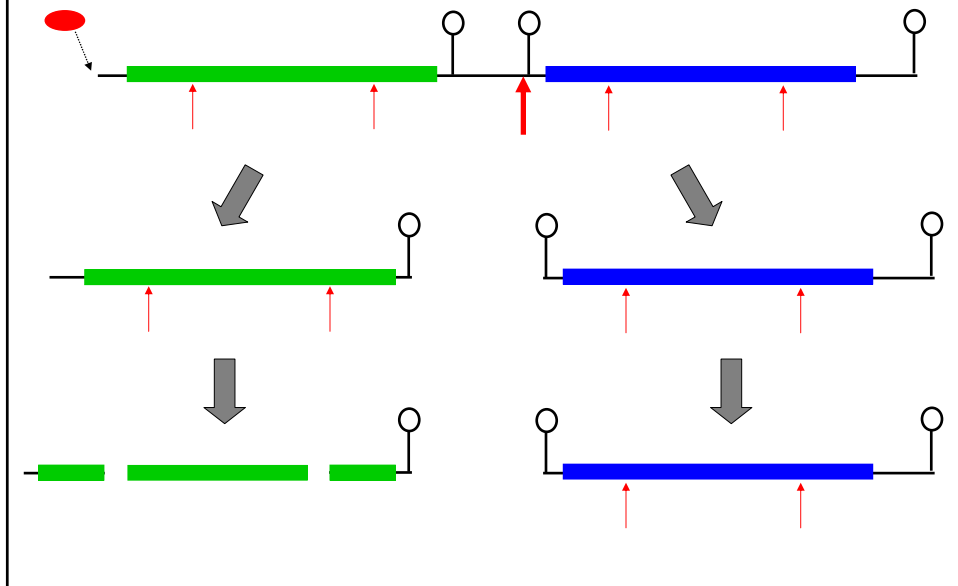
A strategy to design mRNA stability?



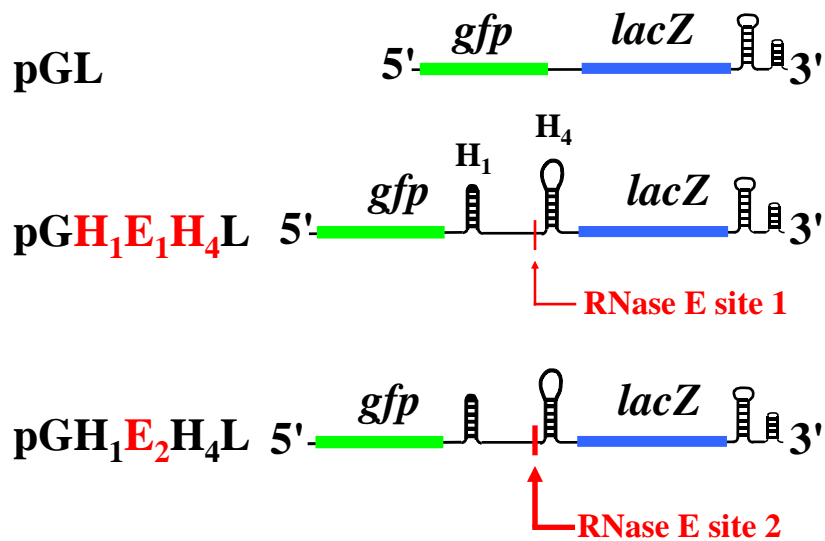
Degradation of multicistronic mRNA

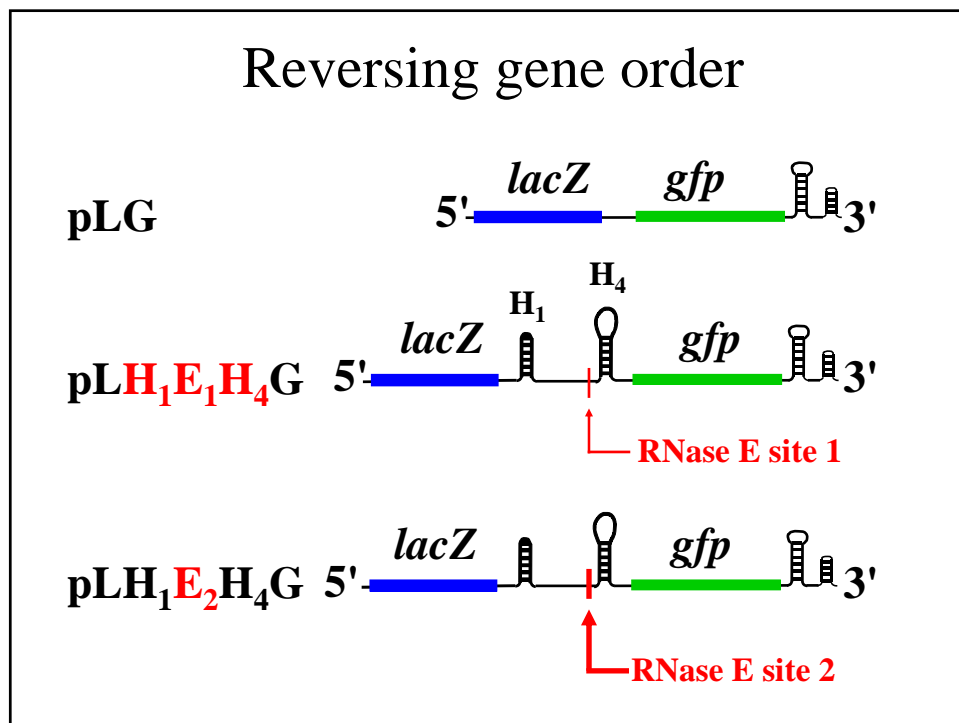
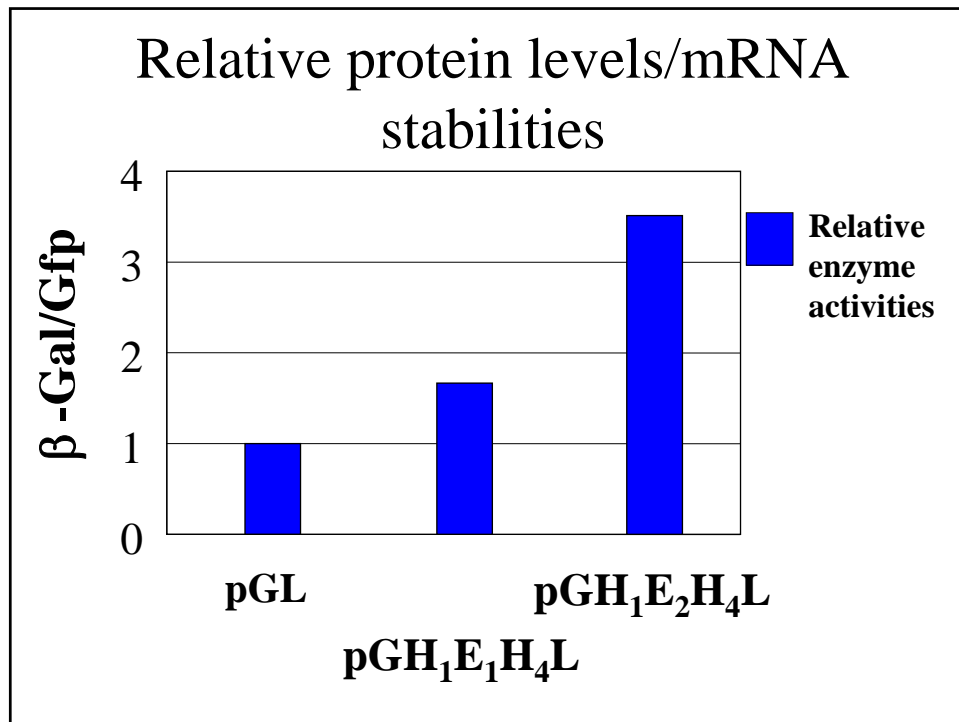


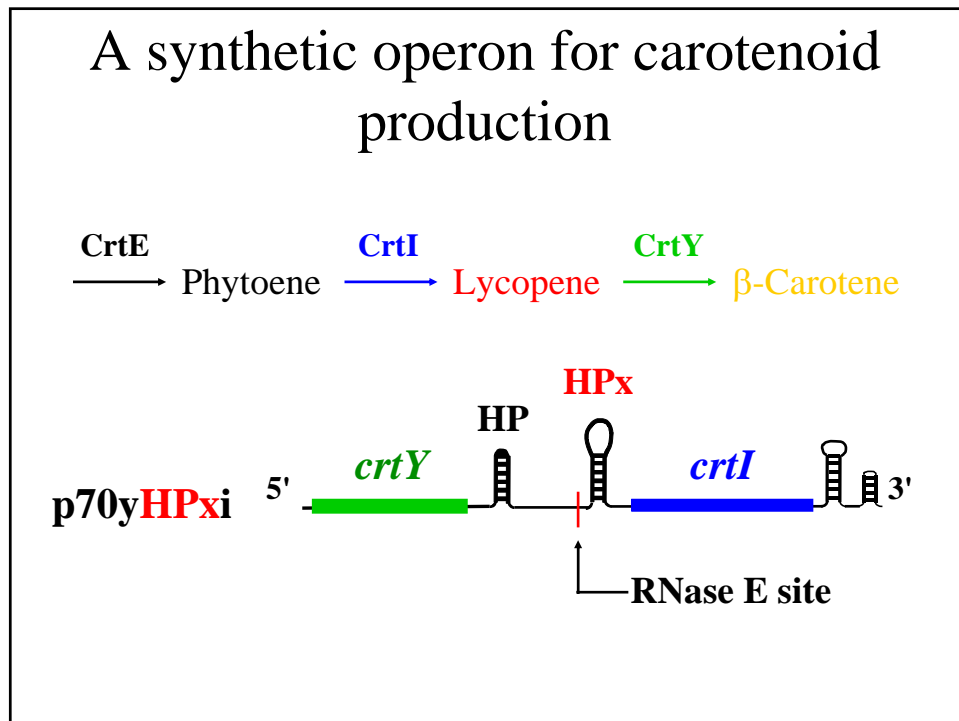
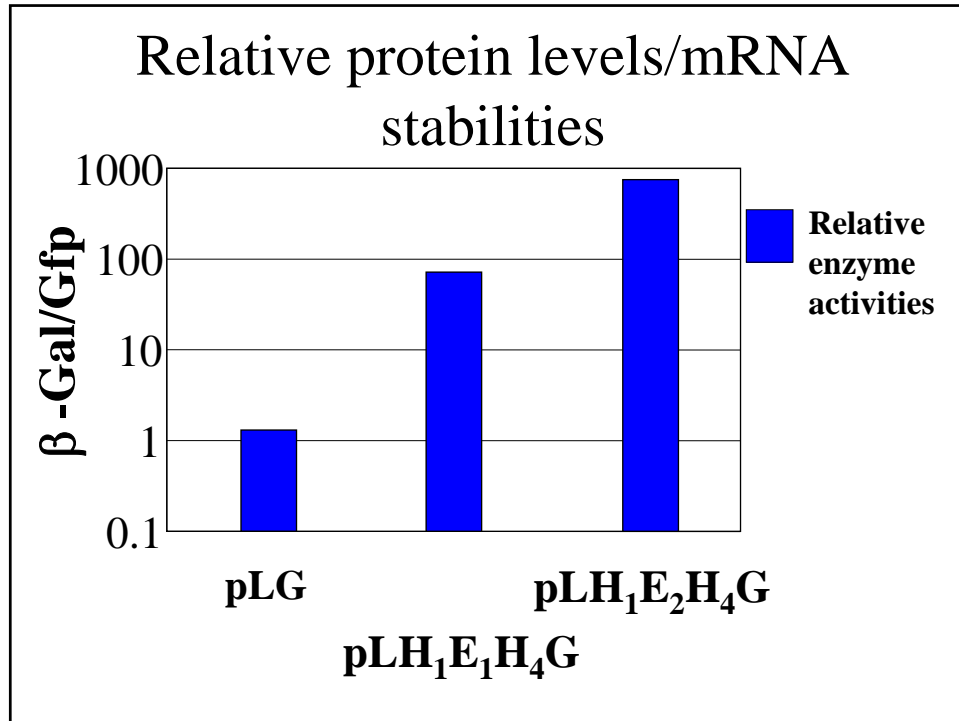
Degradation of multicistronic mRNA



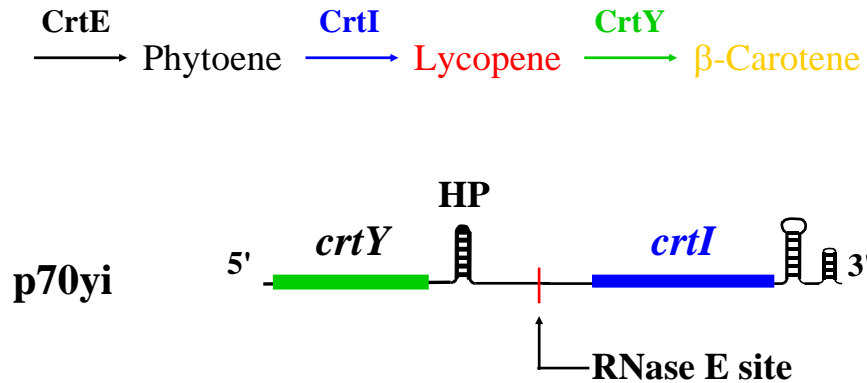
Constructs to test operons



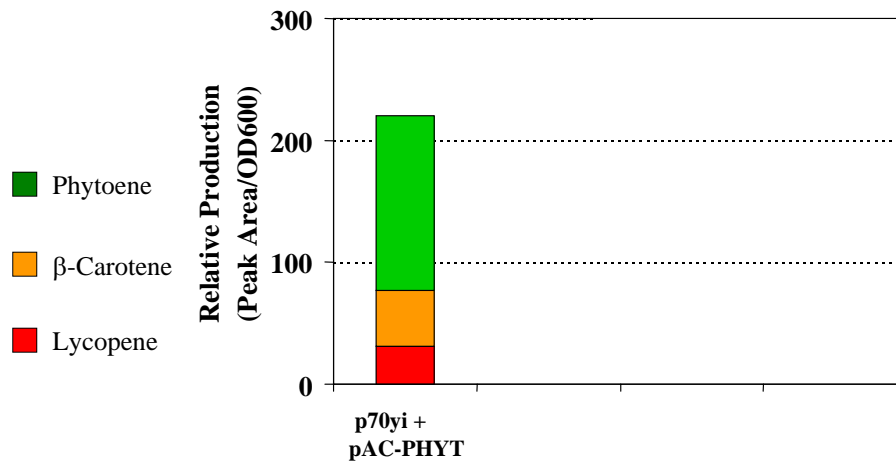


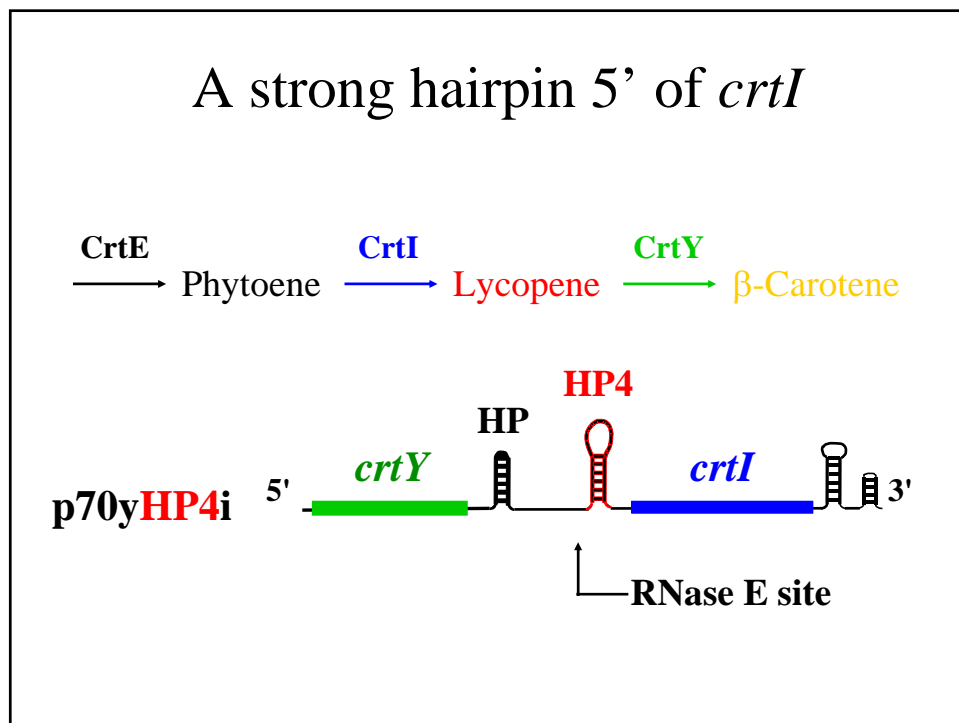
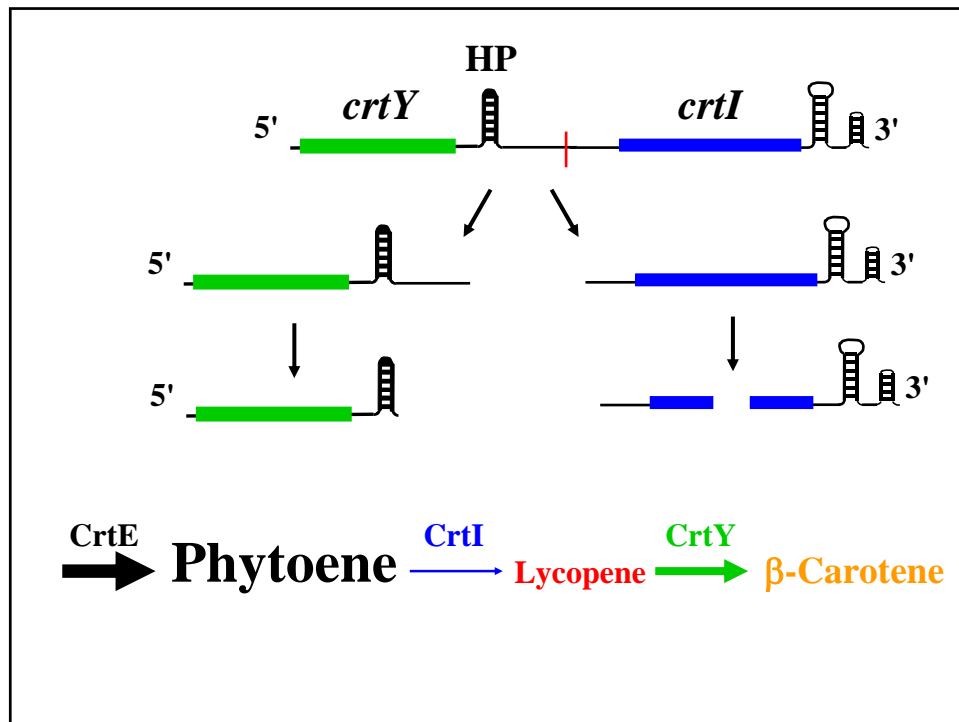


No hairpin 5' of *crtI*

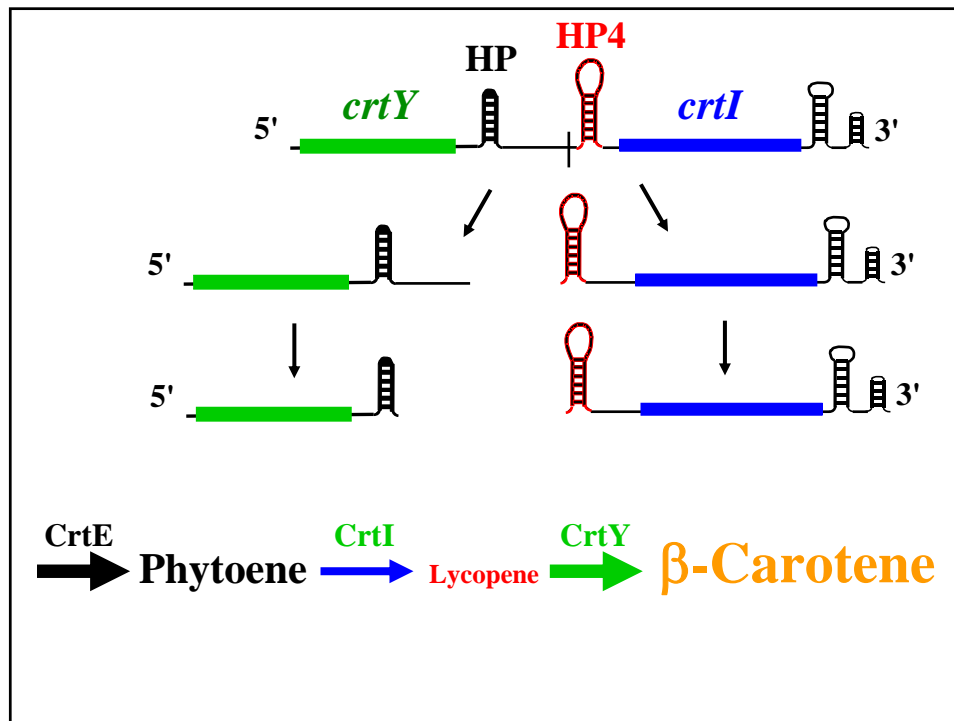
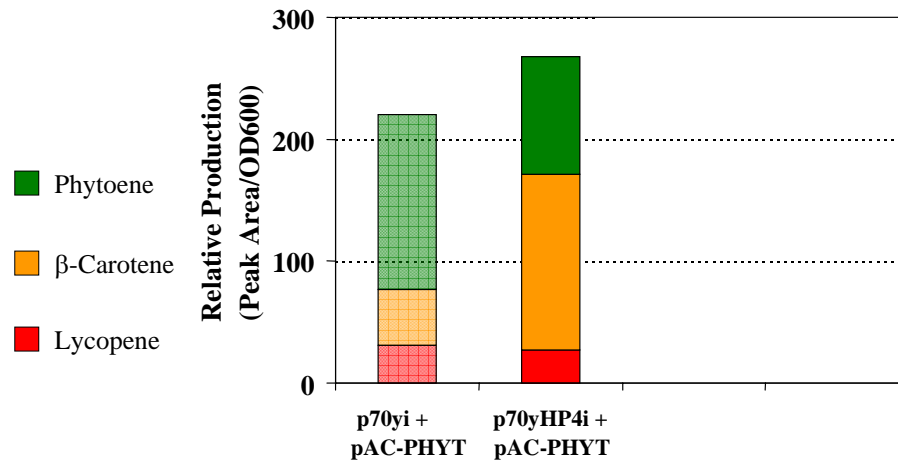


Carotenoid Production

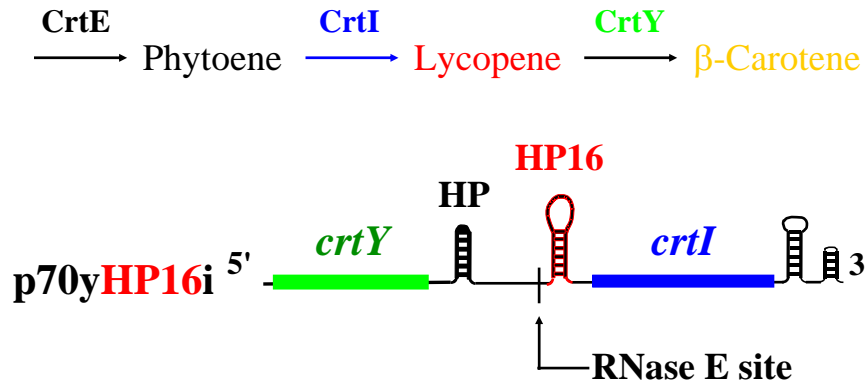




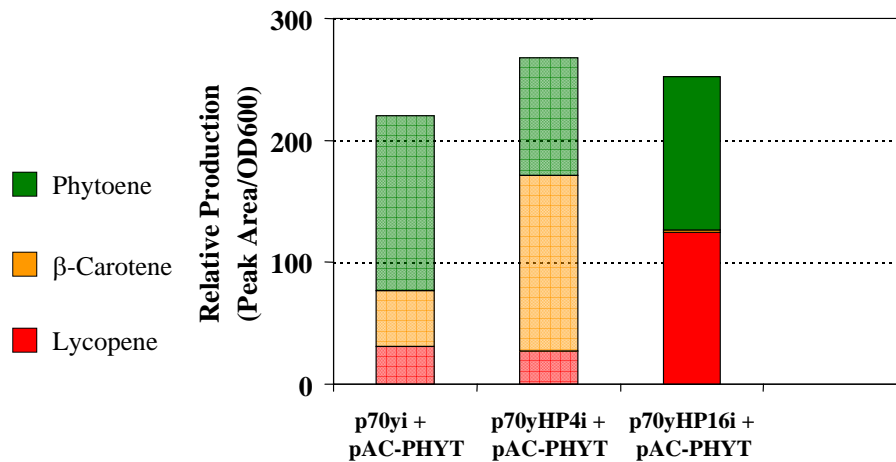
Carotenoid Production

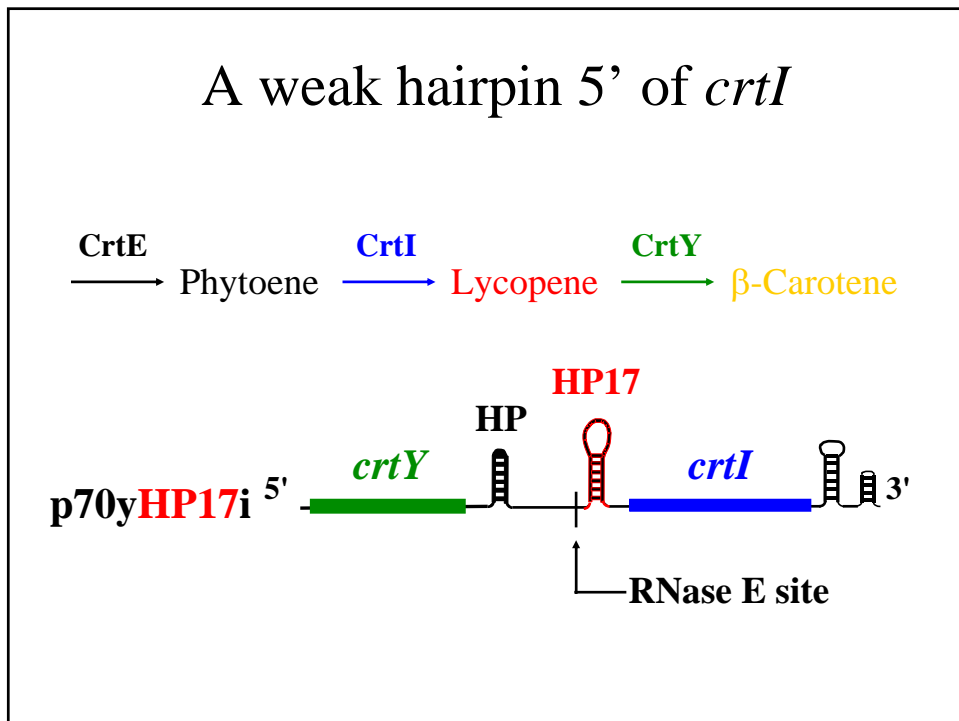
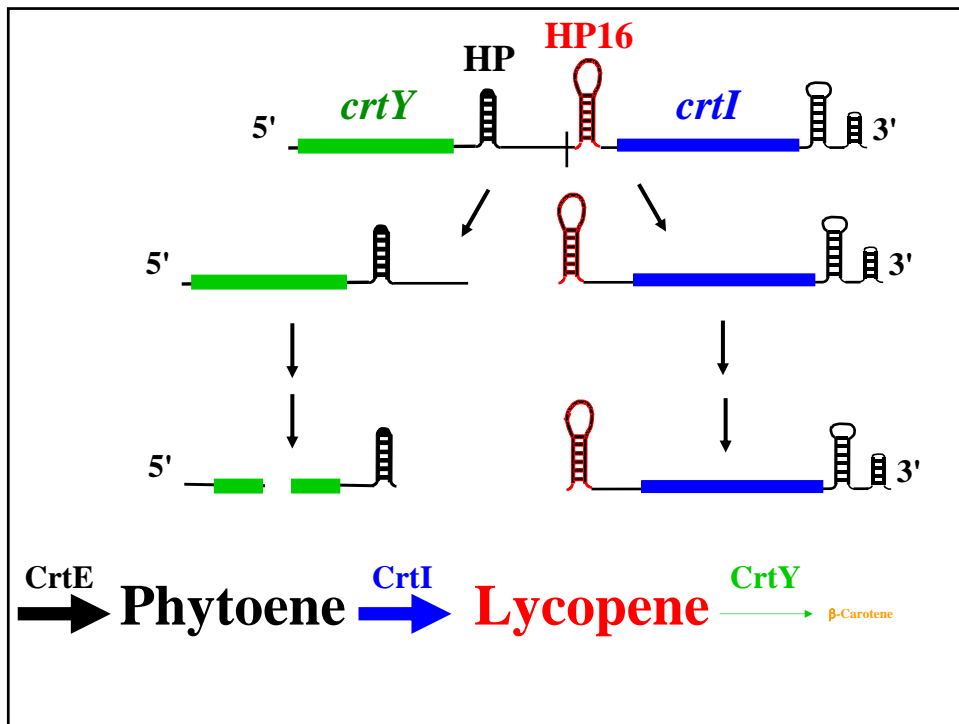


A stronger hairpin 5' of *crtI*

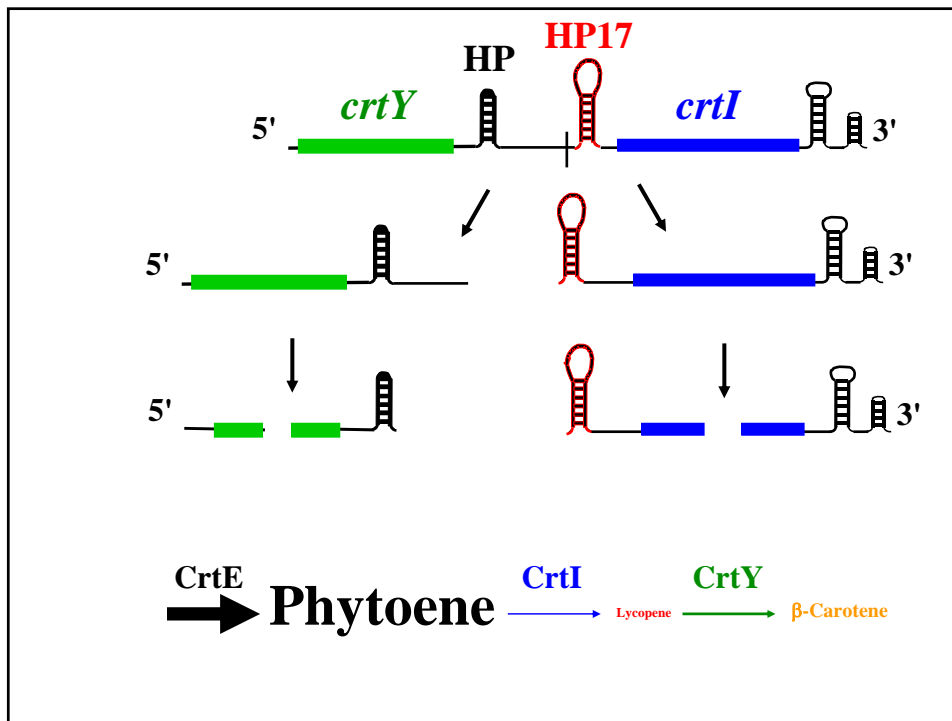
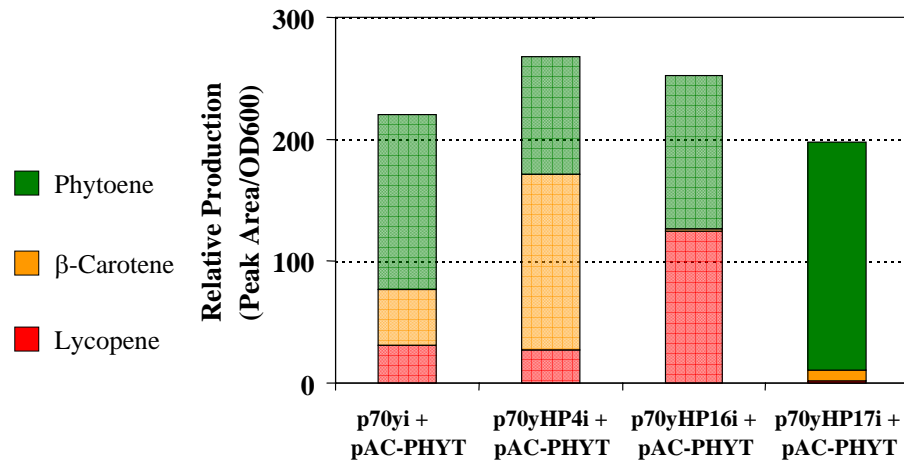


Carotenoid Production





Carotenoid Production



Gene expression tools for metabolic engineering

Metabolic models to quantify flux

Carotenoids ← GGPP ← DMAPP

How do you coordinate the expression of multiple genes?

Mass Balance on Cellular Metabolites

$$\frac{dX}{dt} = S \cdot v - b$$

where

X = Concentration of metabolites

S = Stoichiometric matrix

-> known enzymatic reactions

b = Uptake, secretion, and biomass synthesis

-> known from cell composition

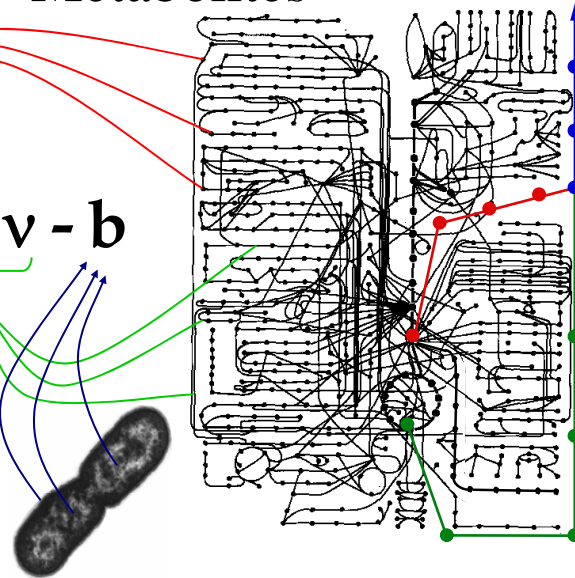
v = Reaction flux vector

-> unknown

Mass Balance on Cellular Metabolites

$$\frac{dX}{dt} = S \cdot v - b$$

Genomics
Proteomics
Physiomics



Solve for fluxes using linear optimization

➔ At steady-state: $S \cdot v = b$

number of
fluxes $>$ number of
metabolites
(495) (289)

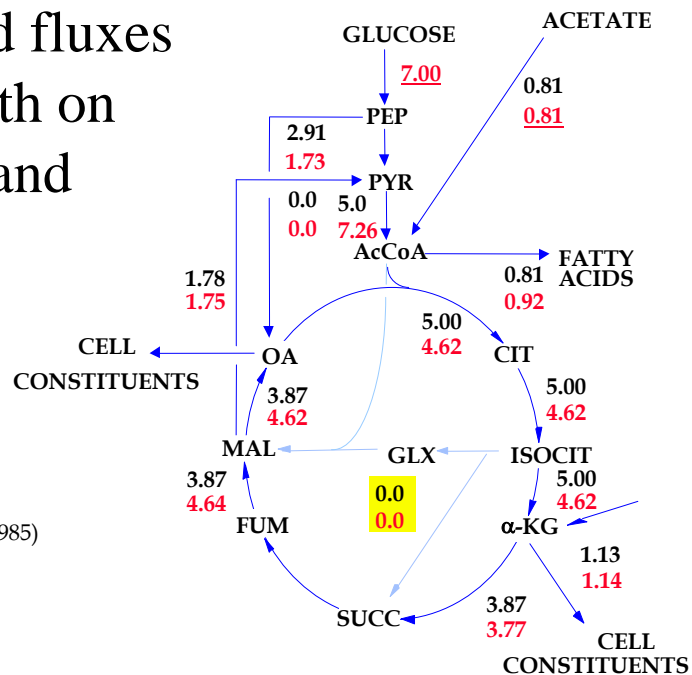
➔ **Linear Optimization:**

Objective Function: $Z = \sum_i c_i \cdot v_i$

Constraints:

1. $S \cdot v = b$
2. Lower Bound $< v_i <$ Upper Bound $i = 1, 2, \dots$

Predicted fluxes
for growth on
glucose and
acetate



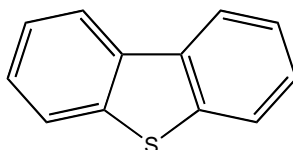
Experimental Data
Walsh & Koshland (1985)

Model Predictions

Some environmental examples of metabolic engineering

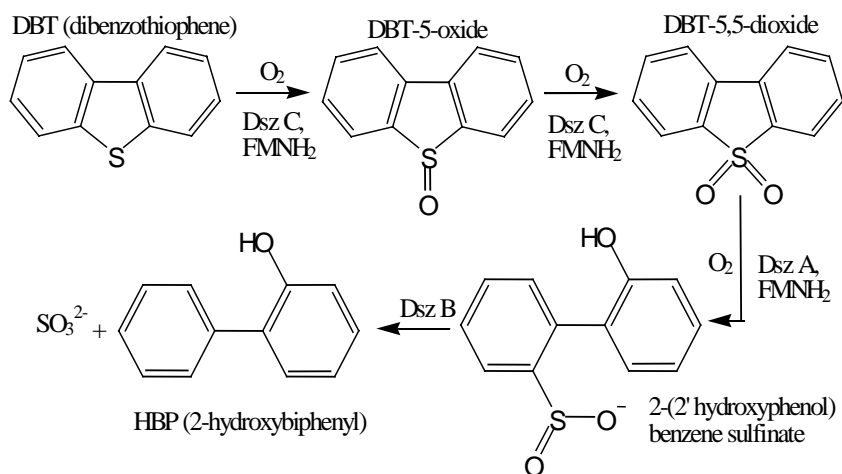
Application of metabolic engineering to biodesulfurization of fossil fuels

- Dibenzothiophene (DBT) is typical of the organic sulfur compounds found in fossil fuels.
- DBT is recalcitrant to hydrodesulfurization.
- Used extensively in biodesulfurization studies

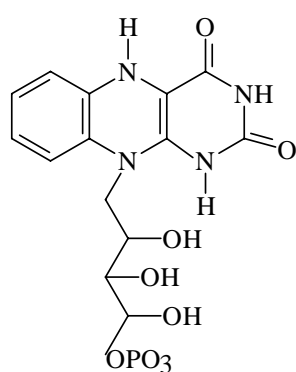


DBT (dibenzothiophene)

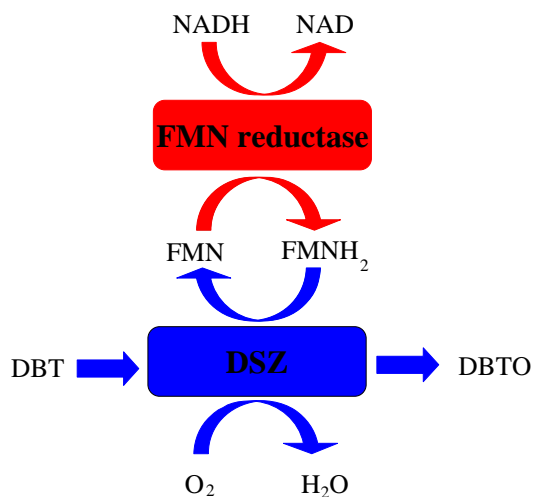
Rhodococcus erythropolis IGTS8 desulfurization pathway



Role of reducing equivalents



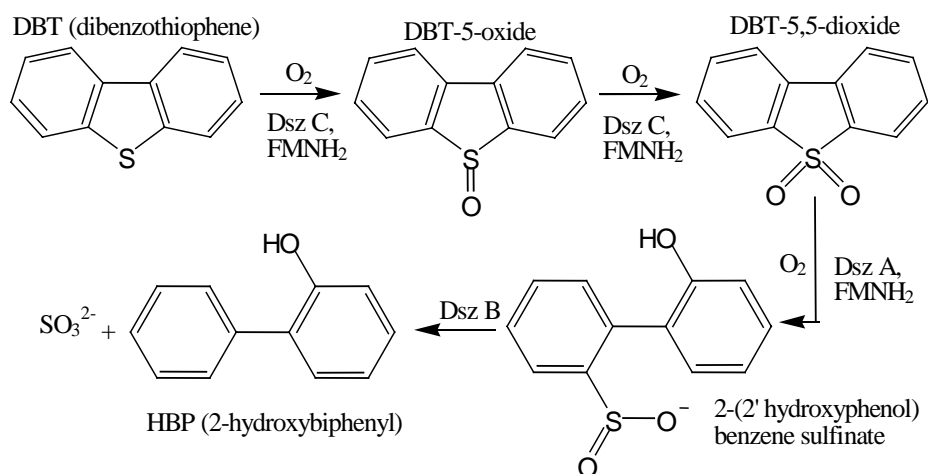
- DszA & DszC require reduced flavin (FMNH_2).



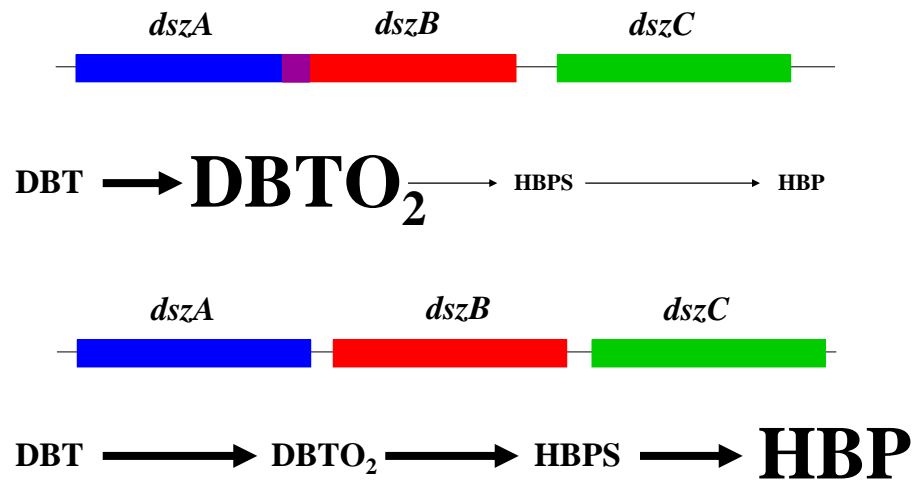
Comparison of degradation rates

Organism	Rate (mg/hr/g dcw)	Source
<i>E. coli</i> pRED/pDSZ	51	This work
<i>Rhodococcus</i> <i>erythropolis</i> IGTS8	3	Gallardo (1997)
<i>Rhodococcus</i> <i>erythropolis</i> H-2	5	Oshiro (1996)

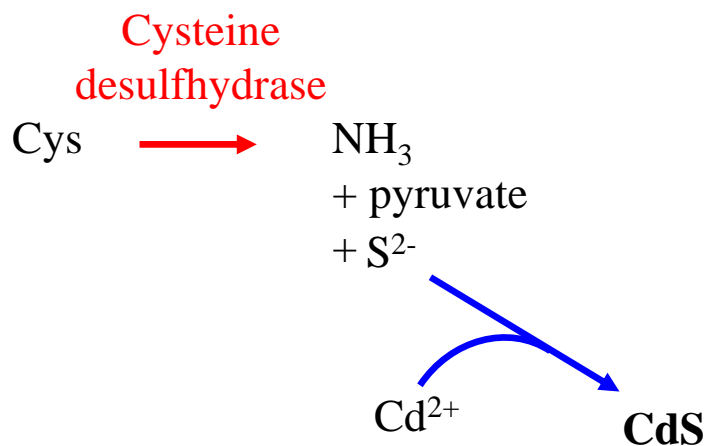
Relieving bottlenecks in the desulfurization pathway



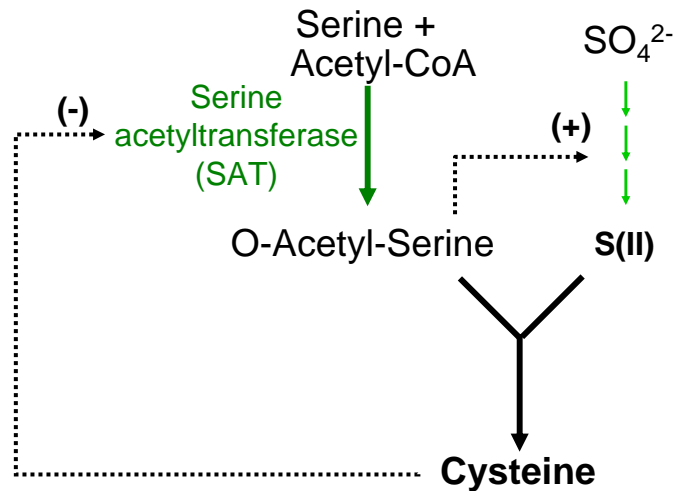
Overlapping reading frames in *dszAB* may limit flux through the pathway



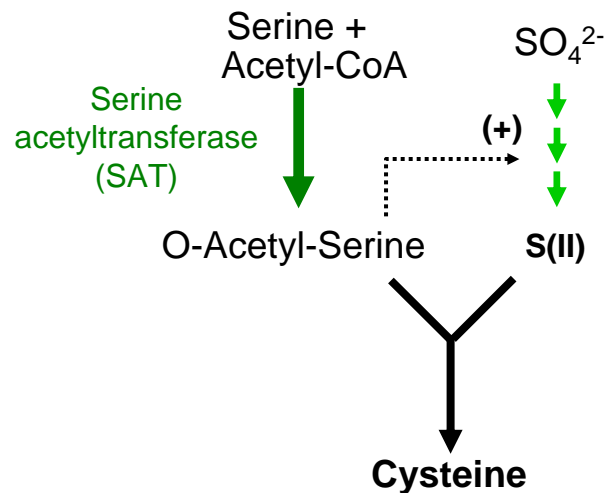
Aerobic precipitation of cadmium



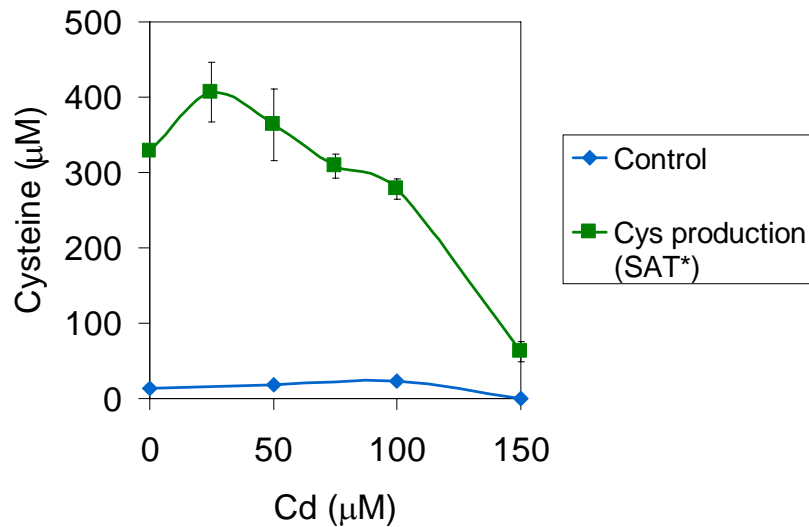
Native cysteine biosynthesis pathway



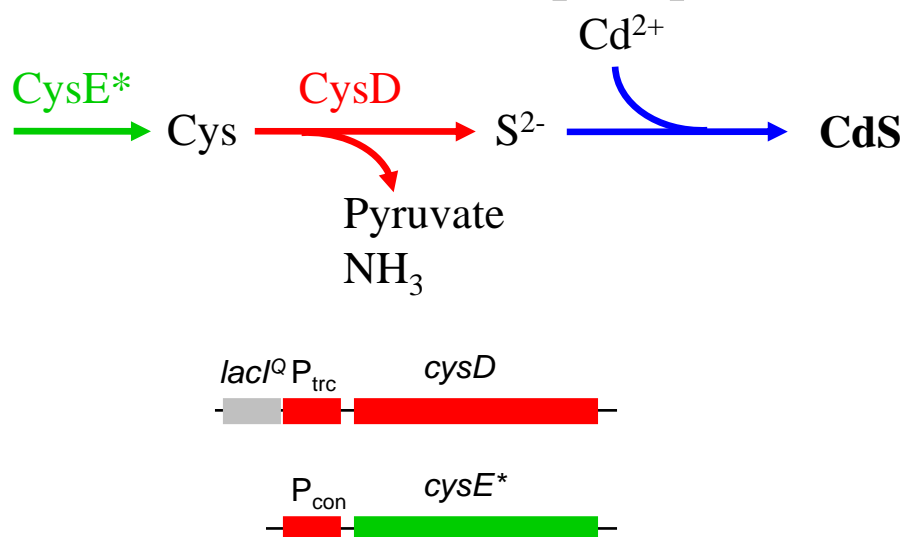
Deregulated cysteine biosynthesis pathway



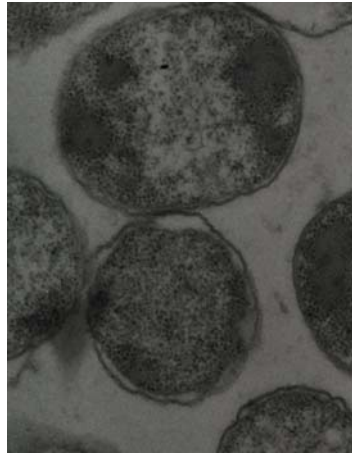
Cysteine production is inhibited by cadmium



Expression of *cysE** and *cysD* for aerobic cadmium precipitation



Precipitation of CdS by the engineered strain

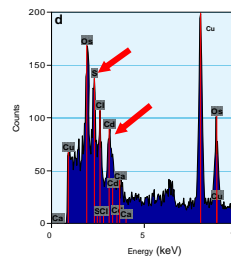
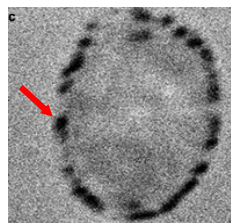
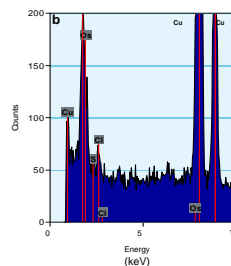
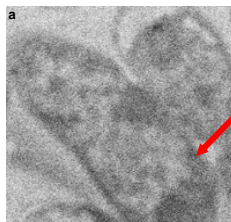


Control



Engineered strain

Cadmium precipitates on the cell wall as cadmium sulfide



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